

# Improvement in Performance and Emission Characteristics of a Single Cylinder S.I. Engine Operated on Blends of CNG and Hydrogen

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**Abstract**—This paper presents the experimental results of a single cylinder Enfield engine using an electronically controlled fuel injection system which was developed to carry out exhaustive tests using neat CNG, and mixtures of hydrogen in compressed natural gas (HCNG) as 0, 5, 10, 15 and 20% by energy. Experiments were performed at 2000 and 2400 rpm with wide open throttle and varying the equivalence ratio. Hydrogen which has fast burning rate, when added to compressed natural gas, enhances its flame propagation rate. The emissions of HC, CO, decreased with increasing percentage of hydrogen but NO<sub>x</sub> was found to increase. The results indicated a marked improvement in the brake thermal efficiency with the increase in percentage of hydrogen added. The improved thermal efficiency was clearly observed to be more in lean region as compared to rich region. This study is expected to reduce vehicular emissions along with increase in thermal efficiency and thus help in reduction of further environmental degradation.

**Keywords**—Hydrogen, CNG, HCNG, Emissions.

## I. INTRODUCTION

MOST of the energy requirement in the world is supplied by fossil fuels. With the increasing concerns over the environmental protection and shortage of fossil fuels, much effort has been focused on the utilization of alternative fuels in engines. Natural gas is thought to be a good alternative to traditional vehicle fuels since it has cleaner combustion characteristics and plentiful reserves.

Natural gas spark ignition (S.I.) engines either run stoichiometrically i.e. exactly enough air for complete combustion or with an excess of air – named lean burn engine. The lean limit is the maximum air-fuel ratio where the engine may run without experiencing misfire. A part from low emissions of NO<sub>x</sub>, lean operation has some other advantages. Firstly, excess air could increase the ratio of specific heats ( $k=C_p/C_v$ ) of the burned gas and improve combustion efficiency, both of which are beneficial to the engine's thermal efficiency [1]; secondly possibilities of knock become smaller since cylinder temperature decreases, thus higher compression ratio which is also good to thermal efficiency could be employed.

However, as the natural gas engine runs close to the so-called lean limit, problem of misfiring occurs. The reason for this is the slow burning speed of natural gas. One effective

method to solve the problem of slow burning velocity of natural gas is to mix the natural gas with the fuel that possesses fast burning velocity. Hydrogen is regarded as the best gaseous candidate for natural gas due to its very fast burning velocity, much better lean-burn capability, and small quenching distance. And this combination is expected to improve the lean burn characteristics [2], [3] and decrease vehicular emissions.

## II. PRIOR RESEARCH

Many researchers have studied the effect of adding hydrogen into natural gas on the performance and combustion characteristics in the past several years.

Akansu et al. [4] experimented on four cylinder engine with mixtures of hydrogen in methane of 0, 10, 20 and 30% by volume. Experimentation was done by varying the equivalence ratio, at 2000 rpm and constant load conditions. The result showed that NO emissions increased while HC, CO and CO<sub>2</sub> emission values decrease and brake thermal efficiency values increased with increasing hydrogen percentage.

Karim et al. [5] theoretically studied the addition of hydrogen on methane combustion characteristics at different spark timings. The results indicated that the addition of hydrogen could increase the flame propagation speed, thus stabilizing the combustion, especially the lean combustion process. These studies showed that the exhaust hydrocarbon (HC), CO, and CO<sub>2</sub> concentrations could be decreased when engine operated on the natural gas engine. However, NO<sub>x</sub> would increase for the natural gas-hydrogen combustion as combustion temperature increased.

Kehraman et al. [6] conducted the experimental study on the performance and exhaust emissions of a spark-ignition engine fuelled with methane – hydrogen mixtures (100% CH<sub>4</sub>, 10% H<sub>2</sub> + 90% CH<sub>4</sub>, 20% H<sub>2</sub> + CH<sub>4</sub>, 30% H<sub>2</sub> + CH<sub>4</sub>) at different engine speeds and different excessive air ratios. Experiments were performed at 1500, 2000, 2500 and 3000 rpm and at wide open throttle (WOT). HC, CO<sub>2</sub> and CO emission values decreased with increasing hydrogen fraction in methane-hydrogen mixtures. Brake thermal efficiencies were found to increase with the increase of H<sub>2</sub> fraction and engine speed.

Ma F et al. [7] conducted experimental research on a spark ignition natural gas engine using variable composition hydrogen/CNG mixtures. They analysed COV in IMEP %, and

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combustion duration/degree crank angle, NO<sub>x</sub>, HC, CO, indicated thermal efficiency by varying excess air ratio. The results showed that lean burn limit could be extended by hydrogen addition. 10%, 30% and 50% hydrogen addition extended lean limit to 1.82, 2.09 and 2.4 respectively compared to 1.71 for natural gas. Hydrogen addition resulted to higher NO<sub>x</sub> emission if spark timing is not optimized. Unburned hydrocarbon emission always decreased with the increase of amount of hydrogen added no matter whether spark timing was changed or not.

Hydrogen enriched compressed natural gas engine has attracted the attention of many researchers. There are many published papers regarding the experimental researches of the performance and emission characteristics hydrogen enriched natural gas engines. Results of these studies generally show that hydrogen addition can decrease engine's unburned hydrocarbon and NO<sub>x</sub> emissions (by lean burn) and speed up the combustion process. More comprehensive experimental study needs to be conducted under a wide range of hydrogen fraction and on commercially available engine so that the research can go from lab to land. For this research, instead of a research engine, commercially available engine was selected and ECU was developed and was tested to evaluate the effect of hydrogen addition on the natural gas engine's emission and performance characteristics at constant speed.

### III. EXPERIMENTAL SETUP TEST PROCEDURE

The experiments were conducted on a modified single cylinder Royal Enfield engine with a bore x stroke of 69.874 x 90 mm and a compression ratio of 7.25:1. The engine details are given in Table I. This engine has been commercially used in the motor bikes in U.K. and India from couple of decades. Fig. 1 is a schematic of the experimental setup.

TABLE I  
ENGINE SPECIFICATIONS

Parameter	Specifications
Engine Make	Bullet (Royal Enfield India Ltd.)
Type	Vertical Single Cylinder, Air Cooled, 4-Stroke with overhead
Bore (mm)	69.874
Stroke (mm)	90
Swept Volume	346 c.c.
Compression Ratio	7.25 : 1
Intake Valve Opening	30° BTDC
Intake Valve Closing	50° ABDC
Exhaust Valve Opening	65° BBDC
Exhaust Valve Closing	15° ATDC

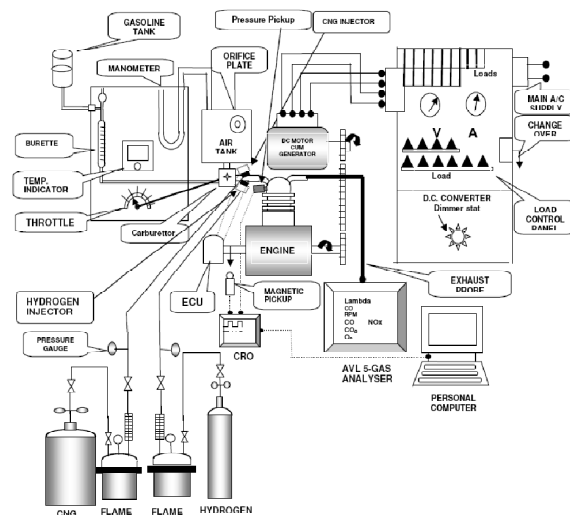


Fig. 1 Schematic of experimental setup

TABLE II  
FUEL PROPERTIES OF NATURAL GAS AND HYDROGEN

Property	Hydrogen	Natural Gas
Density at 1 atm and 300°k (kg/m <sup>3</sup> )	0.082	0.754
Mass lower heating value (MJ/kg)	119.930	43.726
Stoich. Air/ Fuel mass ratio (kg/kg)	34.20	17.19
Volumetric lower heating value at 300°k and 1 atm (MJ/m <sup>3</sup> )	9.82	32.97
Equivalence ratio of lean-burn limit at 293°k and 1 atm	0.1	0.53
Volumetric fraction of fuel at stoichiometric A/F ratio (%)	29.0	9.5
Molar carbon to hydrogen	0	0.25
Quenching gap at NTP (mm)	0.64	2.03
Laminar Flame speed (m/s)	2.90	0.38
Adiabatic Flame temp (°k)	2318	2148
Minimum ignition energy (mJ)	0.02	0.29
Flammability limits (% by volume)	4–75	5.3–15.0
Conductivity at 300°k and 1 atm (mW /m <sup>2</sup> k)	182	34
Octane Number	130+	127

All the research work was conducted in the Engines and Unconventional Fuels Laboratory in the Centre for Energy Studies at the Indian Institute of Technology Delhi. The tests were conducted by varying the equivalence ratio at a constant speed of 2000, and 2400 rpm and constant spark timing of 25° BTDC. The CNG-H<sub>2</sub> supplying system of the engine consisted of CNG and hydrogen tank, pressure regulators, flame traps, Electronic Control Unit (ECU) and two different injectors for injecting CNG and hydrogen. ECU was used to inject the fuel timely and vary the duration of injection and thus vary the

equivalence ratio. The flow rate was metered by Proline Promass 80 A04 micro-motion flow meters which use the Coriolis effect for a direct measure of mass flow. The hydrogen used in this study has a purity of 99.99%. Table II gives the fuel properties of natural gas and hydrogen. It can be seen that the laminar burning velocity of hydrogen is seven times to that of natural gas. Thus, adding hydrogen into natural gas is expected to increase the flame propagation speed and stabilize the combustion process.

In this study, four fractions of natural gas-hydrogen blends were studied. The fractions of hydrogen in the natural gas-hydrogen blends are 0%, 5%, 10%, 15% and 20% by energy respectively.

#### IV. RESULTS AND DISCUSSIONS

##### A. Carbon Monoxide Emissions

Fig. 2 shows the brake specific carbon monoxide as a function of equivalence ratio with the engine speeds of 2000 and 2400 rpm. CO emission is mainly due to incomplete combustion. It is seen that CO emissions are very low in lean region up to equivalence ratio of 0.65. As the mixture is further leaned out, the flame propagation becomes so slow that incomplete combustion takes place and hence CO emission increases sharply. When the equivalence ratio was increased above 0.65, the CO emission reached the minimum and then increased rapidly when the mixture was made further rich. The low emissions of CO occurred in fuel air equivalence range of 0.7 to 0.82. In this range, complete combustion ensured maximum conversion of carbon atoms to  $\text{CO}_2$ . CO values are highly concentrated when equivalence ratio is increased above 0.86 i.e. in fuel-rich mixtures, due to incomplete combustion. As the percentage of hydrogen is increased, CO emissions decrease. This is due to higher combustion temperature of hydrogen which results in more complete combustion. The CO emissions decrease with the increase in the engine speed. The probable reason for this is due to the increase in cylinder temperature as at high engine speed, less time is available for the heat to conduct from cylinder to cylinder walls. The results are consistent with Ref [8].

##### B. Unburned Hydrocarbon Emissions

Fig. 3 shows the brake specific unburned hydrocarbon (HC) versus equivalence ratio. HC emissions are maximum at the leaner mixture, decrease with equivalence ratio and once again increase with rich mixture. HC emissions are reduced by excess air (lean mixture) until reduced flammability of the

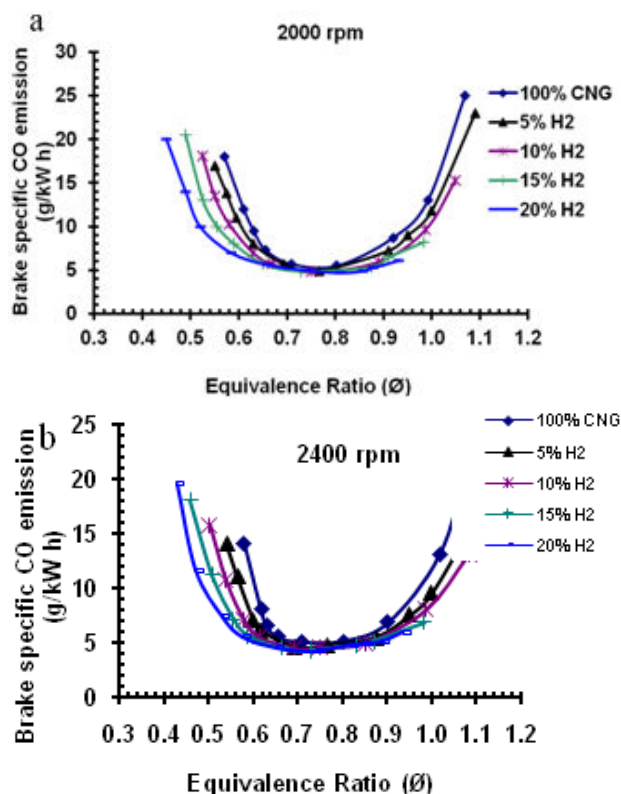
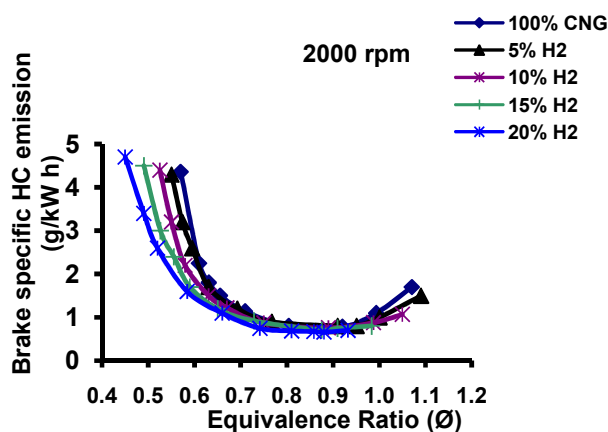


Fig. 2 (a) and (b) variation of brake specific CO emissions versus equivalence ratio for various hydrogen blends at 2000 and 2400 r/min

mixtures causes a net increase in HC emissions. The HC emissions are minimum between equivalence ratios of 0.8 to 0.9. This is because at this point, firstly there was extra air to ensure complete combustion and secondly the fuel-air mixture was not too lean, so the exhaust temperature could keep at high level which was beneficial to the further oxidation of HC



(a)

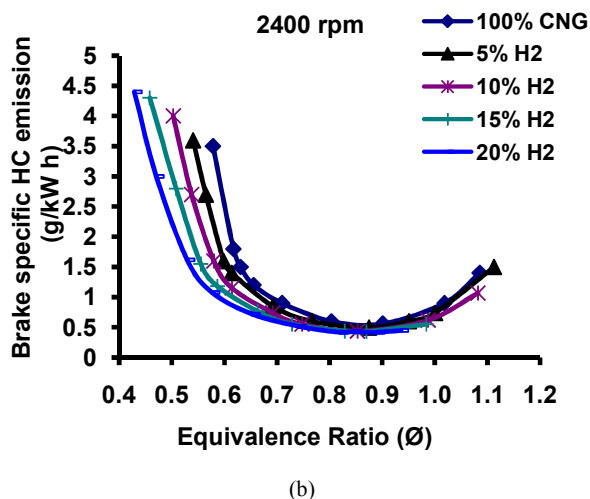


Fig. 3 (a) and (b) variation of brake specific HC emissions versus equivalence ratio for various hydrogen blends at constant speed of 2000 and 2400 r/min

formed through crevice and flame quenching. The emissions increase at high equivalence ratio due to incomplete combustion. With increasing percentage of hydrogen, HC emissions reduced which could be explained by the fact that hydrogen could speed up flame propagation and reduce quenching distance, thus depressing the possibilities of incomplete combustion [9].

Moreover, with the increase in hydrogen percentage, carbon concentration of the fuel blends decreased which resulted in reduced HC emissions. 0.9. This is because at this point, firstly there was extra air to ensure combustion completeness and secondly the fuel-air mixture was not too lean, so the

#### C. NO<sub>x</sub> Emissions

Fig. 4 shows brake specific NO<sub>x</sub> versus equivalence ratio. NO<sub>x</sub> emissions increase with higher temperature, longer high temperature combustion duration and greater availability of oxygen (up to a point). In the present study, brake specific NO<sub>x</sub> reach peak around  $\phi = 0.82$  to 0.9. Higher combustion temperature and availability of sufficient oxygen seem to be the cause of NO<sub>x</sub> levels peaking around the stoichiometric value. However the in-homogeneity in the cylinder shift the maximum brake specific NO<sub>x</sub> towards leaner side of stoichiometric value. For a specific equivalence ratio, brake specific NO<sub>x</sub> concentration increased due to increase of peak combustion temperature by hydrogen addition. The increase in engine speed resulted in increase of brake specific NO<sub>x</sub> emissions. This probably occurred due to increase in cylinder temperature.

#### D. Brake Thermal Efficiency

Fig. 5 shows the BTE versus the equivalence ratio. The brake thermal efficiencies are found to increase with hydrogen addition to CNG. In case of lean mixtures, specific heat values are found to be more reduced than stoichiometric equivalence ratio value. In case of rich mixtures, combustion is incomplete

due to insufficient oxygen. As complete combustion doesn't occur, there will be a decrease of a certain level of thermal efficiency, when equivalence ratio is increased above

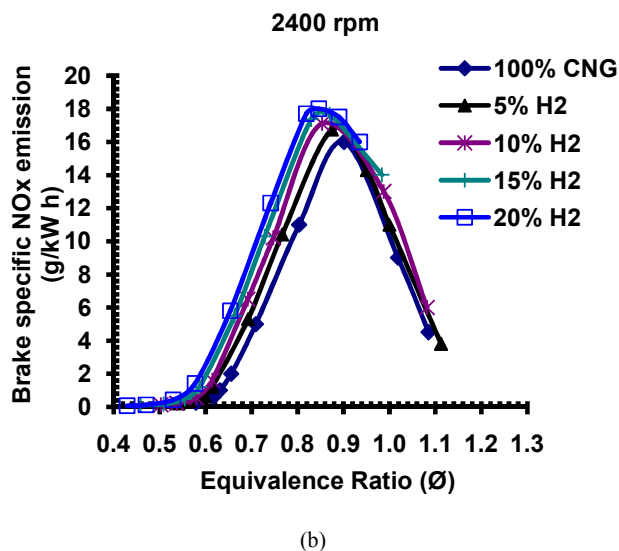
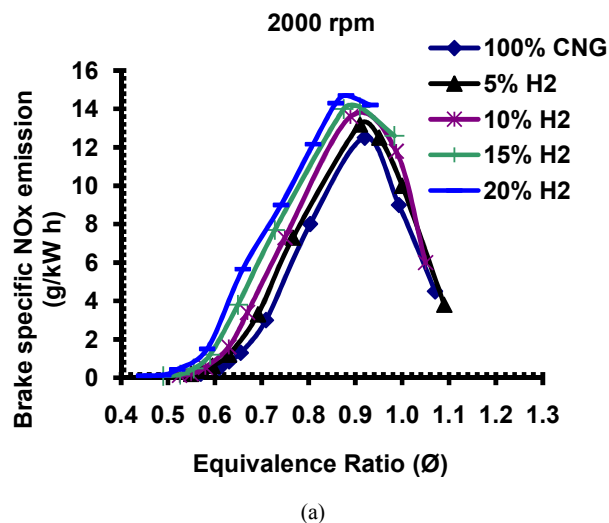
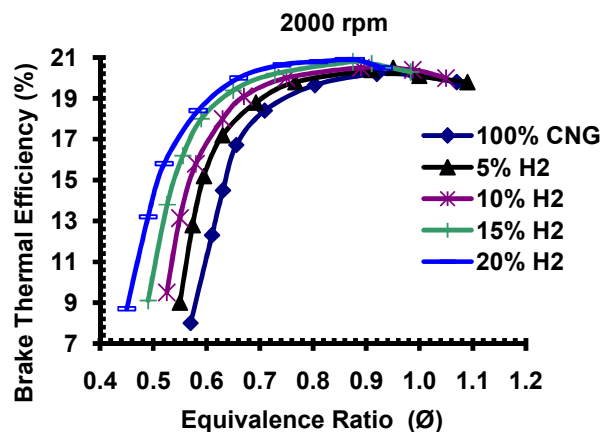
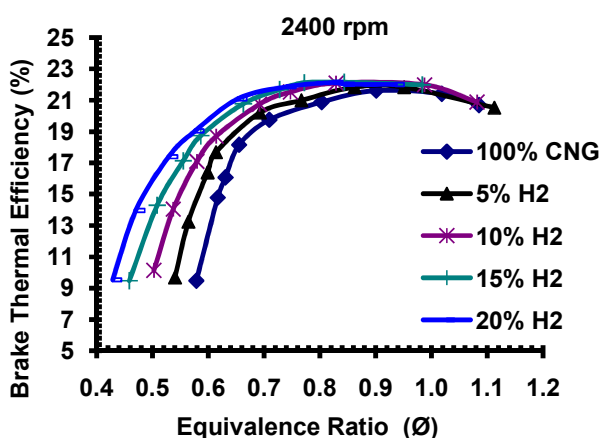


Fig. 4 (a) and (b) variation of brake specific NO<sub>x</sub> emissions as a function of equivalence ratio for various hydrogen blends at constant speed of 2000 and 2400 r/min



(a)



(b)

Fig. 5 (a) and (b) variation of brake thermal efficiency as a function of Equivalence Ratio for various hydrogen blends at constant speed of 2000 and 2400 r/min

stoichiometry. The highest brake thermal efficiency was observed around 0.82 to 0.98 equivalence ratio. In the case of rich mixtures, combustion is not complete. The trend obtained is similar in character to [10].

## V. CONCLUSIONS

An experimental study on the performance, combustion and emission characteristics of a natural gas fuelled engine supplemented with 0%, 5%, 10%, 15% and 20% hydrogen supplementation was conducted. The main results are summarized as follows.

- The timed manifold injection through electronic fuel injection was very advantageous for the smooth operation of the engine.
- The BTE increase with the increasing hydrogen percentage. This increase in BTE is more pronounced in the leaner region.

- The emissions of CO and HC decreased with the increasing percentage of hydrogen.
- The increasing percentage of hydrogen resulted in the increased emissions of NOx. NOx level is very low at lean mixture.

CNG has slower laminar burning velocity and higher ignition energy, both of which have negative effects on the engine's lean burn capability and thus the aim of low CO and NOx can't be realized. At the present stage, the use of neat hydrogen as a fuel in IC engine seems to be a long term prospect mainly due to its undesirable properties and thus the fear attached with its usage. In the initial stage, the addition of hydrogen in small percentages can improve the reputation of hydrogen as a dangerous fuel. The blending of hydrogen in CNG can be very much beneficial for increasing the lean burn limit and for a trade off relation between HC, CO and NOx emission.

The author hope that this study could give some practical guidance to the development and calibration of hydrogen supplemented CNG engines and reduce vehicular emissions and thus decrease further environmental degradation.

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