

Mucus Secretion Responses to Various Sublethal Copper (II) Concentrations in the Mussel *Perna perna*

Kamleshan Pillay

Abstract—The purpose of this study was to evaluate the use of mucus production as a biomarker. This was done by exposing the mussel *Perna perna* to various sublethal concentrations of Cu. Mussels are effective as a bioindicator species as they accumulate Cu in their tissues. Differences in mucus production rates were evaluated at different Cu concentrations. The findings of this study indicate that increasing Cu concentrations had a significant effect on the mucus production rates over a 24 hour exposure. There were also significant differences between the mucus production rates at different Cu concentrations ($p < 0.05$). Thus, mucus is an essential detoxification mechanism.

Keywords—Copper, Mucus, Depuration, *Perna perna*.

I. INTRODUCTION

HEAVY metals can have significant impacts on the physiological, behavioural and cellular responses of organisms if their concentrations are above the critical threshold [16], [19], [23]. Heavy metals may enter the marine environment in a variety of ways. Erosion, wind and volcanic activity are natural pathways that heavy metals may be introduced into coastal systems from [10]. Heavy metals may be introduced via anthropogenic activities such as dumping of industrial wastewater [9],[10], oil spills [24] and sewage effluent [8]. Hence, there is a need to determine the pollution status of marine systems before negative impacts on the ecosystem become irreversible.

The biomarker approach is the use of quantitative variations in processes within an organism in reaction to the exposure of a foreign substance, while bioindicators are changes that occur at higher levels of organisation such as at the population or ecosystem levels [12]. Mussels are an example of a bioindicator species that can be used in establishing the pollution status of an ecosystem [23]. Mussels are ideal bioindicators as they have the ability to accumulate heavy metals from solution and food particles allowing for the pollution status of the area to be ascertained [10], [15], [23]. Hence, the pollution status of the ecosystem can be remediated before the impacts become irreparable. Being sedentary organisms, mussels are easy and inexpensive to sample hence they can be used extensively in marine pollution impact assessments [6], [10].

Copper is a widespread pollutant in the marine environment

[2]. According to [14], copper is a vital metal for bivalve molluscs in low concentrations. Byssus thread formation and metabolism is heavily dependent on low concentrations of copper [14], [20]. However, copper in elevated concentrations can be lethal [14], [15]. Mussels are suitable indicators of high copper concentrations as they may accumulate 3000 times more copper than the surrounding water [22].

Perna perna (brown mussel) is found along the African and South American coasts and in the Gulf of Mexico [1]. *Perna perna* is the most prevalent species of mussel found on the east of South Africa [10]. There has been a steady increase in urban development along the coast of KwaZulu-Natal, South Africa [1]; subsequently, there is a growing need to monitor the pollution status of coastal and estuarine waters. *Perna perna* would be the most suitable bioindicator species since it is indigenous and widespread along the South African coastline [10].

Variations in physiological rates [10], accumulations of heavy metal within tissue [10], [15], [23] and assessing changes in detoxifying secretions [11], [21], [23] in mussels are the most common methods used in determining the pollution status of an area. Mucus secretion by mussels is an example of a detoxifying secretion [11], [21], [23]. Increased production of mucus is evident when high concentrations of heavy metals are present [10], [14], [21], [22], [23]. Hence the evaluation of mucus production rates as a biomarker is necessary. This study aims to determine if differences in mucus production rates between differing copper concentrations are present and if a significant trend exists between increasing copper concentrations and mucus production rates.

II. MATERIALS AND METHODS

A. Sample Collection and Preparation

Samples of *P. perna* were collected from the rocky shore at Park Rynie beach on the south coast of KwaZulu-Natal (30° 18' S; 30° 44' E). The byssus threads of the mussels were cut and all epibionts were removed from the shell. Mussels collected from the field were left to acclimate in an aerated re-circulation tank of 1000 litres for three days in a glasshouse before being moved to the exposure tanks. The temperature was not regulated in the re-circulation tank hence individuals were exposed to fluctuations in the ambient temperature. Mussels were left to acclimate further in aerated filtered

K. Pillay is with the University of KwaZulu-Natal, Durban, KZN, 4000, South Africa (phone: +2726418327; e-mail: pillay.kamleshan@gmail.com).

seawater at 24.4°C within the exposure tanks for 24 hours, prior to the introduction of copper. During the exposure and acclimation within these tanks, individuals were purged. Mussel shell lengths ranged between 32 and 59 mm (mean shell length 44.49 ± 6.01 mm). According to the ANOVA test, there was a no significant difference between size measurements for different Cu concentration treatments ($p > 0.05$, $F = 2.479$, degrees of freedom (D.F.) = 21).

B. Experimental Design

Mussels were placed into four exposure tanks with seven litres of filtered seawater which was aerated. The exposure tanks were an example of a static system. Each exposure tank contained seven mussels. A 500 ml stock solution of distilled water and CuCl_2 [0.74074 g/l , 350 mg/l Cu^{2+}] was prepared. Each of the four tanks contained a different concentration of Cu ($0 \text{ } \mu\text{g/l Cu}$ (control), $12.5 \text{ } \mu\text{g/l Cu}$, $25 \text{ } \mu\text{g/l Cu}$ and $50 \text{ } \mu\text{g/l Cu}$). 1 ml of stock solution was added to the $50 \text{ } \mu\text{g/l Cu}$ concentration. Mussels were exposed for 24 hours.

C. Mucus Collection

Sampling protocol for the extraction of mucus was adapted from [23]. Mussels were taken out of the exposure tanks and placed in filtered seawater for 30 minutes so that may recover from handling. Glass microscope slides were pushed into the mantle cavity of the mussel when the valves of the mussel were opened. Thereafter, microscope slides and mussels were left in pre-weighed glass beakers for 10 minutes. Exposure of mussels to air allows for the collection of mucus [4], [5], [23]. The microscope slide was removed after 10 minutes and mucus washed into the pre-weighed beaker with distilled water. Pre-weighed beakers were placed in an oven at 80°C for 24 hours. Thereafter, the beakers were weighed and the dry mass of mucus was calculated by the difference of the oven dried mucus and beaker and the dry beaker weight.

D. Tissue Collection

After mucus collection, mussels were frozen for 12 hours to allow for the removal of tissue. Shell lengths were recorded in mm using a calliper. Tissue was scraped off from each mussel using a scalpel into pre-weighed petri dishes. Petri dishes were placed in an oven at 80°C for 24 hours. Thereafter, the petri dishes were weighed and the dry tissue mass (g) calculated using the formula of the difference of the oven dried petri dish with dry tissue (g) and the dry petri dish weight (g). Furthermore, the mucus production rates were calculated in grams of mucus per gram of dry tissue per minute by the formula of dividing the mucus in grams by the tissue mass and a time of 10 minutes.

E. Statistical Analysis

Statistical tests were performed using SPSS v 15.0 for Windows while Graphpad Prism 5 was used to generate the graphs. A Grubb's test was performed on all data with 1% of all significant outliers being removed (<http://www.graphpad.com/quickcalcs/Grubbs1.cfm>). A one way ANOVA test with a Bonferroni multi-comparison test

was performed on mucus production rates at different Cu concentrations. A simple linear regression was also performed to investigate if a significant trend exists and the magnitude of the relationship for mucus production rates against increasing concentrations of Cu. All assumption tests for normality and equality of variance were satisfied for all statistical tests ($p > 0.05$). The regression assumption test for linearity was also satisfied.

III. RESULTS

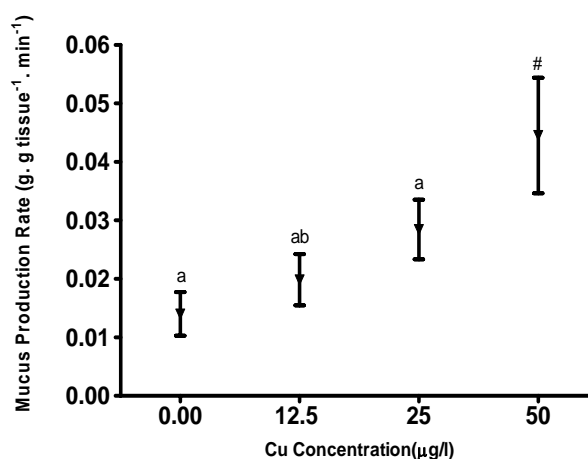


Fig. 1 Mean Mucus production rates with 95% confidence in *P. perna* at different Cu concentrations (same letters indicate that a significant difference between treatments is present, # indicates that the treatment is significantly different from all other treatments)

A significant ANOVA test ($p < 0.005$, $F = 19.679$, $N = 83$) indicates that there was a difference between the mucus production rates at different Cu concentrations. According to Fig. 1, the highest mucus production rate, $0.044 \pm 0.0217 \text{ g.g tissue}^{-1} \cdot \text{min}^{-1}$, occurred at the $50 \text{ } \mu\text{g/l Cu}$ while the lowest mucus production rate was measured at $0.014 \pm 0.0082 \text{ g.g tissue}^{-1} \cdot \text{min}^{-1}$ in the control treatment ($0 \text{ } \mu\text{g/l Cu}$). From the Bonferroni multi-comparison test, there was significant difference between the mucus production rate at $50 \text{ } \mu\text{g/l Cu}$ treatment and all other Cu concentrations ($p < 0.005$). There was also a significant difference between the mucus production rate at the $0 \text{ } \mu\text{g/l Cu}$ treatment and the $25 \text{ } \mu\text{g/l Cu}$ treatment ($p < 0.05$). However, there was no statistically significant difference between both the $12.5 \text{ } \mu\text{g/l Cu}$ and $25 \text{ } \mu\text{g/l Cu}$ treatments and the $0 \text{ } \mu\text{g/l Cu}$ and $12.5 \text{ } \mu\text{g/l Cu}$ treatments ($p > 0.05$). The Bartlett's test for equal variances indicates that there was a significant difference between the variances between treatments ($p < 0.05$).

Fig. 2 illustrates the relationship between Cu concentration and mucus production rates. From the regression analysis, there is a significant increasing trend between mucus production rates and Cu concentrations ($p < 0.005$, $F = 60.130$, $\text{D.F.} = 82$). The magnitude of the model indicates that for every 1 unit in Cu concentration ($\mu\text{g/l}$), there is a 0.001 g.g

tissue⁻¹.min⁻¹ increase in the mucus production rate. According to the R² value of 0.426, 42.6% of the variation in the mucus production rate is explained by the increasing Cu concentration.

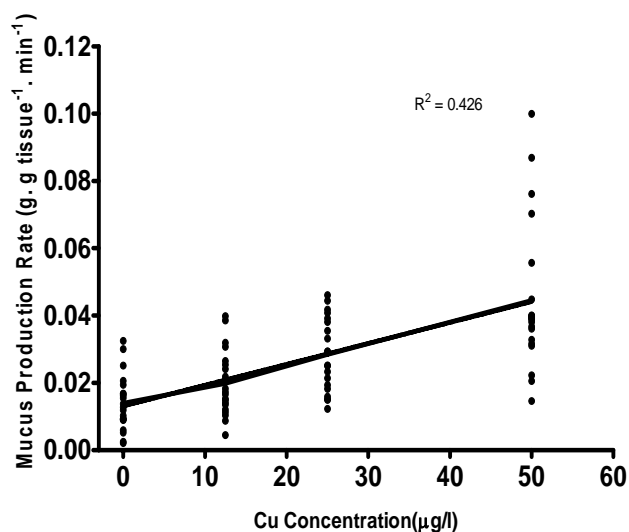


Fig. 2 Mucus production rates when exposed to various concentrations of Cu over 24 hours

IV. DISCUSSION

Mucus can be described as a complex carbohydrate-sulphate [22] which is essential for maintaining homeostasis in all marine molluscs [23]. Mucus is vital in feeding from lining feeding apparatus [7], [23] to selection of particles from solution [7]. It is also integral for locomotion in limpets [3]. In biomonitoring, few studies have been conducted on the role of mucus depuration against environmental pollutants. Mucus is an essential detoxification mechanism [11], [21], [23].

Prior to studies by [21] it was doubted whether there was an actual increase in the mucus production rate when exposed to metals. However the present study and the study by [23] confirm the hypothesis that mucus is fundamental in its role in depuration and indeed increases at higher Cu concentrations during short term exposures (24 hours). The straining of particles from solution is performed by the gills [7]. Mucus may be seen as the first “line of defence” against foreign matter as it may be found across the gills which are constantly in contact with the water [11], [23]. It was found that high concentrations Cu ions would be accumulated in the mucus preventing Cu from building up in the tissues [10], [15], [23]. Hence mucus would be lost back to the water and Cu concentrations in the tissue would be minimal [11], [23]. From the data, higher Cu concentrations result in higher mucus production rates for short term exposures.

Copper is a vital metal in mussels in low concentrations for byssus thread formation and metabolism [14], [20]. Hence mucus production rates not differing in the 0 µg/l Cu concentrations and 12.5 µg/l Cu concentrations are expected.

However, higher concentrations of Cu are toxic [14,15] and subsequently there is a significant difference between mucus production rates in the 0 µg/l Cu and 25 µg/l Cu concentrations as well as between the 0 µg/l Cu and 50 µg/l Cu. According to [11, pg. 103], “Mucus prevents uptake of Cu by binding the positively charged cations onto the active site of mucus-glycoproteins.” Hence mucus prevents Cu ions from building up in the tissue of mussels [10], [15], [23].

[23] and [10] attribute the increase in mucus production rates at increasing Cu concentrations to an increase in the number of mucus glands. However, too high Cu concentrations within tissues are responsible for loss of functionality of mucus glands [23]. Hence it would be expected that the mucus production rate would decrease at higher concentrations. An increasing mucus production rate at higher Cu concentrations could be attributed to the time of exposure. A 24 hour exposure could cause less accumulation of Cu within tissues than the long term study over three months performed by [23]. This is confirmed by the decrease in mucus production rates in the 2nd and 3rd month of exposure by [23] where Cu accumulation in tissue would be much higher, damaging mucus glands. *P. perna* has the ability to depurate Cu from its tissues at a fast rate [10]. Mussels were placed in filtered seawater for 30 minutes to recover from handling before mucus production rates were measured; this could contribute to increasing mucus production rate as Cu would not have sufficient time to damage mucus glands and subsequently decrease mucus production rates.

Perna perna is quite sensitive to high Cu concentrations; it has a 96 hour LC₅₀ of 250 µg/l Cu [13]. Some species of mussel such as *A. trapesialis* have a higher 96 hour LC₅₀ of 2000 µg/l Cu [13]. At the 50 µg/l Cu concentration, it was observed that spawning occurred. Hence, it can be seen that even at this sublethal concentration, mussels are under stress. The Bartlett's test indicates there was significant difference in the variance between different Cu concentrations. This indicates that mussel individuals may have different reactions to high Cu concentrations. Similar individual variation was also observed in cardiac responses [14] and in filtration rates [1].

According to [3], concentrations of metals within mucus may be related to the size of a particular organism. The ANOVA test illustrates that there was no significant difference between the sizes of mussels specimens between treatments hence mussel size was ruled out as a confounding factor. Subsequently, size cannot be taken into account as a parameter that could affect the increasing mucus production rate. Mucus collection was performed after mussels had been fed, which could cause an increase in mucus production rates as mucus is used in particle selection during feeding [7], [11].

The use of mucus production as a biomarker can be recommended [10]. It is used in detoxification [11], [21], [23] and the accumulation of Cu within tissues could be considerable showing a significant increasing relationship with increasing Cu concentration [10], [21], [22], [23]. However, extraneous factors such as temperature, salinity and

water depth need to be taken into consideration [17],[18], [23] as well as the loss of mucus to surrounding water [11],[23]. Feeding could also have an impact on the mucus production rates [7], [11].

In conclusion, it is clear that mucus production rates in mussels show the same increasing relationship during short term exposures ranging from 24 hour to 60 day exposures [23]. However, long term exposures actually cause a decrease in the mucus production rates and these factors need to be taken into account if mucus production rates are to be used as an effective biomarker.

ACKNOWLEDGMENT

I would like to thank my supervisor, Dr. André Vosloo and his wife, Dr. Daléne Vosloo, for their assistance with the experimental design and guidance throughout all phases of this research. Much gratitude is owed to Miss. Prathna Singh and Dr. Joseph Sara for their invaluable help during sampling and specimen collection. Lastly, I would like to thank my colleagues, Mr. Yanasivan Kisten, Mr. Naeem Agjee and Miss. Yashna Maharajh for their dedication and commitment to this research.

REFERENCES

- [1] A. Anandraj, D.J. Marshall, M.A.Gregory, and T.P McClurg, "Metal accumulation, filtration and O₂ uptake rates in the mussel *Perna perna* (Mollusca: Bivalvia) exposed to Hg²⁺, Cu²⁺ and Zn²⁺," Comparative Biochemistry and Physiology Part C: Toxicology & Pharmacology, vol.132, pp. 355-363, July 2002.
- [2] R.J. Brown, T.S. Galloway, D. Lowe, M.A. Browne, A. Dissanayake, M.B. Jones, and M.H. Depledge, "Differential sensitivity of three marine invertebrates to copper assessed using multiple biomarkers," Aquatic Toxicology, vol.66, pp. 267-278, 2004.
- [3] M.S. Davies, and A.M. Hatcher, "Limpet mucus as a depuration route and potential biomonitor," Ecotoxicology, vol. 8, no. 3, pp.177-187, March 1999.
- [4] M.S. Davies, S.J. Hawkins, and H.D. Jones, "Mucus production and physiological energetics in *Patella vulgata* L.," Journal of Molluscan Studies, vol. 56, pp.499-503, 1990.
- [5] M.S. Davies, and G.A. Williams, "Pedal mucus of a tropical limpet, *Cellana grata* (Gould): energetics, production and fate," Journal of Experimental Marine Biology and Ecology, vol. 186, pp. 77-87, 1995.
- [6] M.H. Depledge, A. Aagaard, and P. Györkös, "Assessment of trace metal toxicity using molecular, physiological and behavioural biomarkers," Marine Pollution Bulletin, vol. 31, no. 1-3, pp. 19-27, 1995.
- [7] R.L. Foster-Smith, "The role of mucus in the mechanism of feeding in three filter-feeding bivalves," Journal of Molluscan Studies, vol. 41, pp. 571-588, 1975.
- [8] E. Francioni, A.L.R. Wagener, R.C. Calixto, and G.C. Bastos, "Evaluation of *Perna perna* (Linné, 1758) as a tool to monitoring trace metals contamination in estuarine and coastal waters of Rio de Janeiro, Brazil," Journal of the Brazilian Chemical Society, vol. 15, no.1, pp. 103-110, 2004.
- [9] J.L. Franco, D.B.B. Trivella, R. Trevisan, D.F. Dinslaken, M.R.F. Marques, A.C.D. Baily, and A.L. Dafre, "Antioxidant status and stress proteins in the gills of the brown mussel *Perna perna* exposed to Zinc," Chemo-biological Interactions, vol. 160, pp. 232-240, February 2006.
- [10] M.A. Gregory, D.J. Marshall, R.C. George, A. Anandraj, and T.P. McClurg, "Correlations between metal uptake in the soft tissue of *Perna perna* and gill filament pathology after exposure to mercury," Marine Pollution Bulletin, vol. 45, pp. 114-125, 2002.
- [11] E. Kádár, "Postcapture Depuration of essential metals in the deep sea hydrothermal mussel *Bathymodiolus azoricus*," Bulletin of Environmental Contamination and Toxicology, vol. 78, pp. 45-52, 2007.
- [12] P.K.S., Lam, and J.S. Gray, "The use of biomarkers in environmental monitoring programmes," Marine Pollution Bulletin, vol. 46, pp. 182-186, 2003.
- [13] R. Loayza-Muro, and R. Elías-Letts, "Responses of the mussel *Anodontites trapesialis* (Unionidae) to environmental stressors: Effect of pH, temperature and metals on filtration rate," Environmental Pollution, vol. 149, pp. 209-215, January 2007.
- [14] S. Nicholson, "Cardiac and lysosomal responses to periodic copper in the mussel *Perna viridis* (Bivalvia: Mytilidae)," Marine Pollution Bulletin, vol. 38, no. 12, pp.1157-1162, 1999.
- [15] S. Nicholson, and P.K.S. Lam, "Pollution monitoring in Southeast Asia using biomarkers in the mytilid mussel *Perna viridis* (Mytilidae: Bivalvia)," Environment International, vol. 31, pp. 121-132, May 2004.
- [16] B. Patel and K. Anthony, "Uptake of cadmium in tropical marine lamellibranchs, and effects on physiological behavior," Marine Biology, vol. 108, pp. 457-470, 1991.
- [17] D.J.H. Phillips, "The common mussel *Mytilus edulis* as an indicator of pollution by zinc, cadmium, lead and copper. I. Effects of environmental variables on uptake of metals," Marine Biology, vol. 38, pp. 59-69, 1976a.
- [18] D.J.H., Phillips, "The common mussel *Mytilus edulis* as an indicator of pollution by zinc, cadmium, lead and copper. II. Relationship of metals in the mussel to those discharged by industry," Marine Biology, vol. 38, pp. 71-80, 1976b.
- [19] E. Poulsen, H.U. Riisgaard, and F. Møhlenberg, "Accumulation of cadmium and bioenergetics in the mussel *Mytilus edulis*," Marine Biology, vol. 68, pp. 25-29, 1982.
- [20] W.C. Rose, and M. Bodansky, "Biochemical studies on marine organisms," Journal of Biological Chemistry, vol. 44, pp. 99-112, August 1920.
- [21] D.M. Scott, and C.W. Major, "The effect of copper (II) on survival, respiration, and heart rate in the common blue mussel, *Mytilus edulis*," Biological Bulletin, vol. 143, no. 3 pp. 679-688, December 1972.
- [22] I. Sunila, "Toxicity of Copper and Cadmium to *Mytilus edulis* L. (Bivalvia) in Brackish Water," Annales Zoologici Fennici, vol. 18, pp. 213-223, 1981.
- [23] P.W.C. Sze, and S.Y. Lee, "The potential role of mucus in the depuration of copper from the mussels *Perna viridis* (L.) and *Septifer virgatus* (Wiegmann)," Marine Pollution Bulletin, vol. 31, no.4-12, pp. 390-393, 1995.
- [24] P. Szefer, J. Ge, A.A. Ali, A. Bawazir, and M. Sad, "Distribution and association of trace metals in soft tissue and byssus of mollusc *Perna perna* from the Gulf of Aden, Yemen," Environment International, vol. 23, no. 1, pp. 53-61, 1997.