# Study on Leakage Current Waveforms of Porcelain Insulator due to Various Artificial Pollutants

Waluyo, Parouli M. Pakpahan, Suwarno, and Maman A. Djauhari

Abstract-This paper presents the experimental results of leakage current waveforms which appears on porcelain insulator surface due to existence of artificial pollutants. The tests have been done using the chemical compounds of NaCl, Na<sub>2</sub>SiO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, CaO, Na<sub>2</sub>SO<sub>4</sub>, KCl, Al<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, FeCl<sub>3</sub>, and TiO<sub>2</sub>. The insulator surface was coated with those compounds and dried. Then, it was tested in the chamber where the high voltage was applied. Using correspondence analysis, the result indicated that the fundamental harmonic of leakage current was very close to the applied voltage and third harmonic leakage current was close to the yielded leakage current amplitude. The first harmonic power was correlated to first harmonic amplitude of leakage current, and third harmonic power was close to third harmonic one. The chemical compounds of H<sub>2</sub>SO<sub>4</sub> and Na<sub>2</sub>SiO<sub>3</sub> affected to the power factor of around 70%. Both are the most conductive, due to the power factor drastically increase among the chemical compounds.

*Keywords*—Chemical compound, harmonic, porcelain insulator, leakage current.

## I. INTRODUCTION

**O**VERHEAD transmission or distribution lines are widely used in present power system to transmit electric power from generation stations to customer points. Their proper function depends to a large extent on the insulation system with the supporting structures [1]. The performance of outdoor insulators is affected by some parameters. One of these parameters is environmental contamination. To approach the real condition of pollutant effects on an insulator surface, it has been conducted leakage current measurement of outdoor insulator with the artificial pollutants experimentally. However, the compounds of pollutants theirselves were based on the elements from the coastal region site, i.e. Na, Mg, Al,

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Suwarno is academic staff at School of Electrical Engineering and Informatics, ITB, Indonesia and member of IEEE (e-mail: suwarno@ieee.org). Maman A. Djauhari is Academic staff at Department of Mathematics, ITB, Indonesia (e-mail: maman@math.itb.ac.id). Si, S, Cl, K, Ca, Ti and Fe [2]. For this moment, based on such elements, tests have been done using the compounds of Na<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, Al<sub>2</sub>SO<sub>4</sub>, Na<sub>2</sub>SiO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, KCl, NaCl, CaO, TiO<sub>2</sub> and FeCl<sub>3</sub>.

The objective of this research is to obtain the significance of correlation among parameters. The main parameters to be analyzed are leakage current amplitude related to its first, third, fifth and seventh harmonic amplitudes and powers. Also to determine which chemical compound(s) influence significantly to the leakage current using statistical correspondence and correlation analyses.

## II. EXPERIMENTAL AND ANALYSIS METHODS

The chemical compounds were dissolved or emulsified in the fresh water, and sprayed on the porcelain insulator surface overlapped with one by one of compound emulsions. The porcelain insulator was dried under sun ray, and then put in the chamber, with the humidity and temperature were in room condition. It was subjected to high voltage, where the voltage and leakage current waves were recorded by a two-channel digital storage oscilloscope. The recorded data were transferred to pc for further analysis. The experimental setup schematic diagram is shown on Fig. 1.

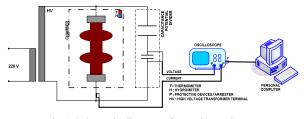


Fig. 1 Schematic diagram of experimental setup

For analysis, it has been done using correspondence analysis with basic rule as follows [3]:

$$P = -\frac{1}{n}K$$
 (1)

where K is the row data matrix, and n is the sum of total components of K. V is the matrix of P divided by the sum of each column and U is the matrix of P, which is divided by the sum of each row.  $D_k$  is the matrix with the diagonal components; sum of each row of P and  $D_m$  is the matrix with the diagonal components; sum of each column of P. Their

non-diagonal components are zero. Furthermore, the analysis on  $R_{\rm k}$  follows

$$T_1 = V D_m V^t \tag{2}$$

$$T_1 D_k^{-1} \vec{u}_k = \lambda_i \, \vec{u}_i, i = 1, 2, \dots, q \tag{3}$$

and

$$\vec{\xi}_i = V_t D_k^{-1} \vec{u}_i \tag{4}$$

and analysis on  $R^{\rm m}$  will result in

ī

$$S = U D_k U^{*}$$
(5)

$$SD_{m}^{T}v_{i} = \lambda_{i}v_{i}, i = 1, 2, ..., q$$
 (6)

$$\vec{d}_i = U_t D_m^{-1} \vec{v}_i \tag{7}$$

Analysis on both Rk and Rm simultaneously are:

$$d_{ij} = \sum_{l=1}^{m} \frac{1}{\sqrt{\lambda_i}} \frac{P_{jl}}{P_{j\bullet}} c_{il}; j = 1, 2, \dots, k$$
(8)

$$c_{ij} = \sum_{l=1}^{k} \frac{1}{\sqrt{\lambda_i}} \frac{P_{ij}}{P_{\bullet,i}} d_{il}; j = 1, 2, \dots, m$$
(9)

It is also analyzed using correlation matrix base on covariance matrix to understand how much level of correlation among parameters base on these research with basic formulae as follows [4]-[5]

$$Cov(X,Y) = \frac{1}{n} \sum_{j=1}^{n} (x_j - \mu_x) (y_j - \mu_y)$$
(10)

Thus, the components of correlation matrix as below.

$$\rho_{x,y} = \frac{Cov(X,Y)}{\sigma_{x},\sigma_{y}} \tag{11}$$

It was also used fast Fourier transform for analysis the leakage current waveforms [6]. Therefore the frequency spectra were obtained. These implementations used the **Danielson-Lanczos** method [7]. The FFT of the sampled data is:

$$V[k] = \sum_{n=0}^{N-1} w[n] x[n] \exp(-i2\pi F_k n), \ 0 \le n \le N-1$$
(12)

and the estimate of the power spectrum is:

$$P[k] = \frac{1}{\sum_{n=0}^{N-1} (w[n])^2} |V[k]|^2$$
(13)

## III. RESULT DATA AND ANALYSIS

Fig. 2 shows the applied voltage and leakage current waveforms of insulator due to 1 gr NaCl solution. It is shown that the leakage current waveform is not pure sinusoidal, i.e. on the peaks, this experiences deformed decrease. This phenomenon indicates that the wave has harmonics. The phase difference between the leakage current and applied voltage waves is 63.4 degree.

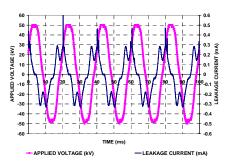


Fig. 2 The waveforms due to 1 gr NaCl solution

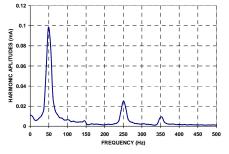


Fig. 3 The leakage current spectrum due to 1 gr NaCl

Thus, it was analyzed in some harmonic frequencies of leakage current waveform, as shown on Figure 3. The highest amplitude, of course, is the fundamental, followed by  $5^{th}$  harmonic, 25.55%, and  $7^{th}$  harmonic, 9.35%. This is also presented the power of harmonics, as shown on Figure 4. The highest is also the fundamental, and followed by  $5^{th}$  and  $7^{th}$  harmonics respectively, or 6.53% and 0.87% of the fundamental.

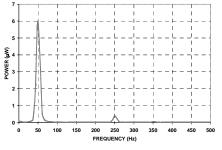


Fig. 4 Power of leakage current spectrum due to 1 gr NaCl solution

The second coating was 1 gr Na<sub>2</sub>SiO<sub>3</sub>. The leakage current waveform was still similar with previous one. It had 58.3 degree of phase difference to the applied voltage. The highest amplitude of harmonic frequency spectrum was the fundamental, and 5.77%, 19.96% and 6.10% of the fundamental for  $3^{rd}$ ,  $5^{th}$  and  $7^{th}$  harmonics respectively. The highest of harmonic power was 11.1  $\mu$ W, and  $3^{rd}$ ,  $5^{th}$  and  $7^{th}$  harmonics were 0.33%, 4.00% and 0.37% respectively.

Fig. 5 shows the applied voltage and leakage current waveforms of insulator due to 1 gr  $H_2SO_4$  solution. The leakage current waveform is strictly different from previous

ones, only one peak on the wave. The phase difference is smaller than the formers, 34.6 degree. This is also shown that the leakage current amplitude is very high, for low applied voltage. For moment, these phenomena indicate that  $H_2SO_4$  solution is very conductive.

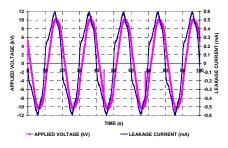


Fig. 5 The waveforms due to 1 gr H<sub>2</sub>SO<sub>4</sub> addition

Fig. 6 shows the harmonic frequency spectrum of leakage current waveform. After the fundamental frequency, the second highest amplitude is  $3^{rd}$  harmonic, 11.12%, and followed by  $5^{th}$ ,  $7^{th}$  and  $9^{th}$ , those are 8.09%, 3.64% and 1.06% of the fundamental respectively.

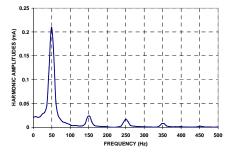


Fig. 6 The leakage current spectrum due to 1 gr H<sub>2</sub>SO<sub>4</sub>

Whereas Fig. 7 shows the power of harmonic frequency spectrum of leakage current waveform. On this condition, the fundamental power is 26.8,  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup>, and 7<sup>th</sup> harmonic powers are 1.24%, 0.65% and 0.13% of the fundamental.

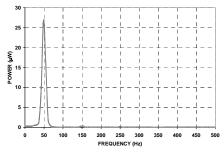
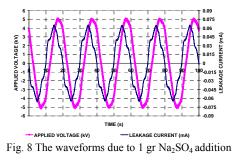


Fig. 7 Leakage current power spectrum of 1 gr H<sub>2</sub>SO<sub>4</sub>

Whereas, on the addition of 1 gr CaO, the leakage current waveform were still similar with  $Na_2SiO_3$  one. The leakage current wave had the phase difference of 62.6 degree. After

the fundamental frequency, the second highest amplitude was 5<sup>th</sup> harmonic, 20.16%, and followed by 3<sup>rd</sup> and 7<sup>th</sup>, those were 6.41% and 4.63% of the fundamental respectively. On this condition, the fundamental power was 9.62  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers were 0.41%, 4.06%, 0.21%, of the fundamental.



Furthermore, Fig. 8 shows the applied voltage and leakage current waveforms of insulator due to addition of 1 gr  $Na_2SO_4$  on previous coating. The leakage current waveform tends to similar to that of  $H_2SO_4$  coating, rather than the others. The leakage current wave has 46.1 degree of phase difference. It is smaller rather than that of CaO, and larger rather than that of  $H_2SO_4$ .

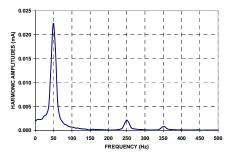


Fig. 9 The leakage current spectrum of 1 gr Na<sub>2</sub>SO<sub>4</sub>

Fig. 9 shows the harmonic frequency spectrum of leakage current waveform. It is indicated that after the fundamental frequency, the second highest amplitude is  $5^{\text{th}}$  harmonic, 9.64%, and followed by  $7^{\text{th}}$  and  $3^{\text{rd}}$ , those are 3.40% and 0.94% of the fundamental respectively.

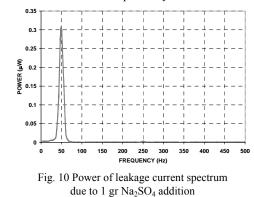


Fig. 10 shows the power of harmonic frequency spectrum of leakage current waveform due to 1 gr Na<sub>2</sub>SO<sub>4</sub>. On this condition, the fundamental power is 0.306  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers are 0.000319  $\mu$ W or 0.01%, 0.92% and 0.10% of the fundamental.

Otherwise, the leakage current waveform due to KCl was not so significantly different from that CaO coating. The leakage current wave had 56.2 degree of phase difference from the applied voltage. It was larger than that of  $Na_2SO_4$  or  $H_2SO_4$ .

Whereas, the harmonic frequency spectrum of leakage current waveform indicated that after the fundamental frequency, the second highest amplitude was 5<sup>th</sup> harmonic, 21.54% and followed by 3<sup>rd</sup> and 7<sup>th</sup>, those were 5.24% and 4.60% of the fundamental respectively. On this condition, the fundamental power of harmonic frequency spectrum of leakage current waveform due to KCl was 8.47  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers were 0.28%, 4.64% and 0.21% of the fundamental.

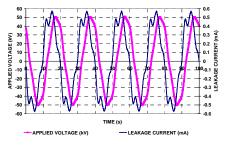


Fig. 11 The waveforms due to 1 gr Al<sub>2</sub>SO<sub>4</sub> addition

Fig. 11 shows the applied voltage and leakage current waveforms of insulator due to addition of 1 gr  $Al_2SO_4$  on previous coating. The leakage current waveform is similar to that of NaCl or Na<sub>2</sub>SiO<sub>3</sub> coating. The leakage current wave has considerable phase difference from the applied voltage, 44.6 degree, and smaller than that of KCl.

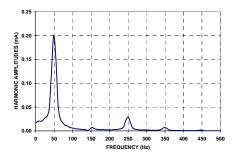


Fig. 12 The leakage current spectrum due to 1 gr Al<sub>2</sub>SO<sub>4</sub> addition

Whereas, the harmonic frequency spectrum of leakage current waveform due to 1 gr  $Al_2SO_4$  addition is shown on Fig. 12. After the fundamental frequency, the second highest amplitude is 5<sup>th</sup> harmonic, 14.48% and followed by 3<sup>rd</sup> and 7<sup>th</sup>, those are 3.57% and 3.47% of the fundamental respectively.

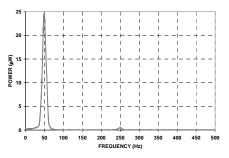


Fig. 13 Power of leakage current spectrum due to 1 gr Al<sub>2</sub>SO<sub>4</sub> addition

The power of harmonic frequency spectrum of leakage current waveform due to  $Al_2SO_4$  is shown on Fig. 13 above. On this applied voltage, the fundamental power is 2.46  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers are 1.27%, 20.93% and 1.20% of the fundamental respectively.

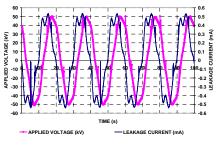


Fig. 14 The waveforms due to 1 gr MgSO<sub>4</sub> addition

Fig. 14 shows the applied voltage and leakage current waveforms of insulator due to addition of 1 gr MgSO<sub>4</sub> on previous coating. The leakage current waveform is similar to  $Al_2SO_4$  coating. The leakage current wave has considerable phase difference, 48.2 degree.

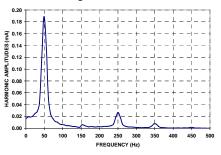


Fig. 15 The leakage current spectrum due to 1 gr MgSO<sub>4</sub> addition

The harmonic frequency spectrum of leakage current waveform due to 1 gr MgSO<sub>4</sub> addition is shown on Fig. 15. After the fundamental frequency, the second highest amplitude is  $5^{th}$  harmonic, 14.24% and followed by  $7^{th}$  and  $3^{rd}$ , those are 3.99% and 3.16% of the fundamental.

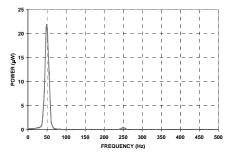


Fig. 16 Power of leakage current spectrum due to 1 gr MgSO<sub>4</sub> addition

Whereas the power of harmonic frequency spectrum of leakage current waveform due to MgSO<sub>4</sub> is shown on Fig. 16 above. On this applied voltage, the fundamental power is 2.18  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers are 1.00%, 20.28% and 1.59% of the fundamental.

The leakage current waveform due to 1 gr FeCl<sub>3</sub> is similar to MgSO<sub>4</sub> coating. The phase angle is also nearly same as 47.5 degree. Whereas the second highest amplitude of leakage current waveform is 5<sup>th</sup> harmonic, i.e. 13% and followed by 3<sup>rd</sup> and 7<sup>th</sup>, those are 4.38% and 3.35% of the fundamental amplitude respectively. The fundamental power of leakage current harmonic frequency spectrum due to FeCl<sub>3</sub> is 7.5  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers are 1.91%, 16.93% and 0.74% of the fundamental respectively.

The leakage current waveform due to 1 gr TiO<sub>2</sub> is similar to that of FeCl<sub>3</sub> coating. Its phase angle is also nearly same as that one, 47.5 degree. After the fundamental frequency, the second highest amplitude is 5<sup>th</sup> harmonic, 13.84% and followed by 3<sup>rd</sup> and 7<sup>th</sup>, those are 5.91% and 3.59% of the fundamental amplitude respectively. Whereas the fundamental power of leakage current harmonic spectrum is 156  $\mu$ W, and the 3<sup>rd</sup>, 5<sup>th</sup> and 7<sup>th</sup> harmonic powers are 0.35%, 1.92%, and 0.13% of the fundamental amplitude respectively. These harmonics are very small comparing to other ones.

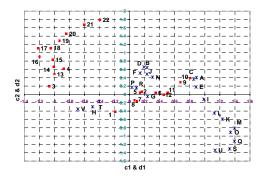


Fig. 17 Correspondence analysis result of experimental data

Fig. 17 shows the scatter plot of correspondence analysis of experimental data. To make shorter of variable names, it is used capital letter.  $V_{max}$ ,  $I_{max}$ ,  $\phi$ ,  $\cos(\phi)$ ,  $1^{st}_{max}$ ,  $3^{rd}_{max}$ ,  $5^{th}_{max}$ ,  $7^{th}_{max}$ ,  $1^{st}_{p}$ ,  $3^{rd}_{p}$ ,  $5^{th}_{p}$ ,  $7^{th}_{p}$ , NaCl, Na<sub>2</sub>SiO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, CaO, Na<sub>2</sub>SO<sub>4</sub>, KCl, Al<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, FeCl<sub>3</sub>, TiO<sub>2</sub> are signed by A, B, until V.

However, individuals or components using numeral indicate different quantities, i.e. on each condition; there are two values, small and large values.

From this scatter plot, it can be obtained some information of experimental data. It is shown components 1 until 12, except 3 and 4, are correlated closely to variable A up to J. This case is indicated that those plots close each other.

For closer view, amplitudes of applied voltages are close to leakage current, phase, power factor, harmonic amplitude and harmonic power. Thus, these quantities are influenced significantly by applied voltage magnitudes.

The amplitude of applied voltage is very correlated with first harmonic amplitude of leakage current. Thus, the first harmonic amplitude increases as the applied voltage to insulator under test rise very significantly. Besides that, it is also influenced by chemical compound of CaO.

From these data, the third harmonic amplitude of leakage current is very close to amplitude of yielded leakage current. Thus, such third harmonic amplitude is influenced by the leakage current amplitude significantly. On rather far from applied voltage amplitude or yielded leakage current, it is shown the fifth harmonic amplitude of leakage current. Therefore, it is rather correlated or influenced by applied voltage or yielded leakage current. The fifth harmonic amplitude of leakage current. The fifth harmonic amplitude of leakage current is also influenced by Na<sub>2</sub>SiO<sub>3</sub> tightly. The latest is seventh harmonic amplitude that is very far from either applied voltage or yielded leakage current. On other word, it is not influenced by either applied voltage or leakage current significantly. However, it is correlated to chemical compounds of MgSO<sub>4</sub> and TiO<sub>2</sub> closely.

The fundamental or first harmonic power is very close to first harmonic amplitude of leakage current. Therefore, both quantities are influenced each other. The first harmonic amplitude rises make the first harmonic power increases, and vice versa.

The third harmonic power is very close to the yielded amplitude of leakage current and third harmonic amplitude of leakage current. Therefore, such three quantities are influenced each other. The yielded leakage current influences third harmonic amplitude and third harmonic power very significantly.

However, the fifth and seventh harmonic powers are very far from either fifth or seventh harmonic parameter. Therefore, both quantities can be said that they are influenced by those harmonic amplitudes slightly. On other hand, both harmonic powers are close to chemical compounds of NaCl,  $H_2SO_4$ ,  $Na_2SO_4$ ,  $Al_2SO_4$  and fairly of FeCl<sub>3</sub>. Thus, it is concluded that the fifth and seventh harmonic powers are influenced by those chemical compounds.

Table I shows correlation matrix that correlate among variables, those have been normalized, so that the correlation between its own self is unity as maximum. The columns from the left to the right, and the rows from the top to the bottom are  $V_{max}$  (maximum applied voltage),  $I_{max}$  (maximum leakage current), phi (phase angle), cos-phi (power factor),  $1_{max}^{st}$ ,  $3_{max}^{rd}$ ,  $5_{max}^{th}$ ,  $7_{max}^{th}$  (maximum leakage current frequency

spectrum harmonics amplitudes 1<sup>st</sup> to 7<sup>th</sup>), 1<sup>st</sup>p, 3<sup>rd</sup>p, 5<sup>th</sup>p, 7<sup>th</sup>p, (maximum power of leakage current frequency spectrum 1<sup>st</sup> to 7<sup>th</sup>), NaCl, Na<sub>2</sub>SiO<sub>3</sub>, H<sub>2</sub>SO<sub>4</sub>, CaO, Na<sub>2</sub>SO<sub>4</sub>, KCl, Al<sub>2</sub>SO<sub>4</sub>, MgSO<sub>4</sub>, FeCl<sub>3</sub> and TiO<sub>2</sub> (kind of chemical compounds for insulator surface coating).

 TABLE I

 CORRELATION MATRIX AMONG THE PARAMETERS

		COR	<b>KELAII</b>	IN IVIA		AMONO	INCI	AKAMI	LIEKS		
	$V_{\rm max}$	$\mathbf{I}_{\mathbf{n}\mathbf{x}}$	phi	Cos(phi)	$l_{\max}^{\prime\prime}$	3 <sup>rd</sup> max	5 <sup>rk</sup> max	7 <sup>m</sup>	$1_p^{st}$	$3_P^{rd}$	5 <sup>#</sup>
Vmax	1	.69	22	.25	.67	.54	.83	.79	.47	.30	.58
$I_{max}$	.69	1	42	.40	.99	.92	.96	.96	.93	.83	.92
phi	22	42	1	99	42	43	33	35	32	37	27
Cos(phi)	.25	.40	99	1	.39	.38	.33	.34	.29	.32	.26
1 <sup>st</sup> max	.67	.99	42	.39	1	.92	.95	.95	.93	.84	.92
311	.54	.92	43	.38	.92	1	.84	.88	.86	.95	.82
5 <sup>m</sup>	.83	.96	33	.33	.95	.84	1	.97	.87	.73	.91
7 <sup>sh</sup> max	.79	.96	35	.34	.95	.88	.97	1	.85	.76	.88
1 <sup>51</sup> <sub>P</sub>	.47	.93	32	.29	.93	.86	.87	.85	1	.89	.98
3 <sup>rd</sup>	.30	.83	37	.32	.84	.95	.73	.76	.89	1	.83
5¢	.58	.92	27	.26	.92	.82	.91	.88	.98	.83	1
7# 7#	.56	.90	28	.27	.90	.85	.89	.90	.95	.88	.97
NaCl	03	.13	43	.41	.13	.07	.07	.10	.12	.10	.09
Naci Na SiO	03	.13	43	.41	.13	.07	.07	.10	.12	.10	.09
H <sub>2</sub> SO <sub>4</sub>	09	.20	71	.70	.22	.16	.09	.05	.10	.20	.14
CaO	.04	.13	59	.61	.13	02	.10	03	.20	.02	.19
Na,SO,	02	.17	58	.57	.18	0	.11	.01	.25	.07	.22
KCI	.14	.30	42	.43	.31	.13	.26	.15	.33	.15	.32
Al,SO4	.11	.37	43	.43	.38	.19	.29	.23	.41	.22	.37
MgSO4	.09	.40	31	.31	.41	.26	.32	.28	.46	.30	.42
FeCl <sub>3</sub>	.07	.47	26	.26	.48	.38	.39	.35	.57	.43	.53
TiO											
110	.05	.42	16	.16	.43	.41	.38	.36	.57	.53	.57
110	.05 7%	.42 NaCl	•.16 Na,SiO,	.16 H,SO4	.43 CaO	.41 Na <sub>2</sub> SO4	.38 KCI	.36 Al,SO4	.57 MqSO∡	.53 FeCl,	.57 TiO
Vmax			•.16 Na <sub>2</sub> SiO, •.06			.41 Na <sub>2</sub> SO <sub>4</sub> .02					
	7%	NaCl	Na <sub>2</sub> SiO <sub>3</sub>	H <sub>2</sub> SO <sub>4</sub>	CaO	Na2504	KCI	Al <sub>2</sub> SO <sub>4</sub>	MgSO4	FeCl <sub>3</sub>	TiO
V <sub>max</sub> I <sub>max</sub> phi	7) .56 .90 .28	NaCl •.03 .13 •.43	Na2SiO3 06 .20 68	H₂SO₄ •.09 .22 •.71	CaO .04 .13 .59	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58	KCI .14 .30 42	Al <sub>2</sub> SO <sub>4</sub> .11 .37 43	MgSO <sub>4</sub> .09 .40 .31	FeCl <sub>3</sub> .07 .47 .26	TiO 0.5 .42 16
V <sub>max</sub> I <sub>max</sub> phi Cos(phi)	7 <u>%</u> .56 .90	NaCl •.03 .13	Na2SiO3 06 .20	H <sub>2</sub> SO <sub>4</sub> 09 .22	CaO .04 .13	Na2SO4 02 .17	КСІ .14 .30	Al <sub>2</sub> SO <sub>4</sub> .11 .37	MgSO <sub>4</sub> .09 .40	FeCl <sub>3</sub> .07 .47	TiO 0.5 .42
V <sub>mex</sub> I <sub>mex</sub> phi Cos(phi) 1 <sup>st</sup> <sub>mex</sub>	7) .56 .90 .28	NaCl •.03 .13 •.43	Na2SiO3 06 .20 68 .70 .20	H₂SO₄ •.09 .22 •.71	Ca0 .04 .13 .59 .61 .13	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58 .57 .18	KCI .14 .30 42	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38	MgSO <sub>4</sub> .09 .40 .31	FeCl <sub>3</sub> .07 .47 .26 .26 .48	TiO 0.5 .42 .16 .16 .43
V <sub>max</sub> I <sub>max</sub> phi Cos(phi) 1 <sup>nt</sup> <sub>max</sub> 3 <sup>rd</sup> <sub>max</sub>	7% .56 .90 .28 .27 .90 .85	NaCl 03 .13 43 .41 .13 .07	Na2SiO3 06 .20 68 .70 .20 .14	H <sub>2</sub> SO <sub>4</sub> 09 22 71 .70 .22 .16	Ca0 .04 .13 .59 .61 .13 .02	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58 .57 .18 0	KCI .14 .30 .42 .43 .31 .13	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19	MgSO <sub>4</sub> .09 .40 .31 .31 .41 .26	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38	TiO 0.5 .42 .16 .43 .43
V <sub>max</sub> I <sub>max</sub> phi Cos(phi) 1 <sup>st</sup> <sub>max</sub> 3 <sup>rd</sup> <sub>max</sub> 5 <sup>th</sup> <sub>max</sub>	7% .56 .90 .28 .27 .90	NaCl .03 .13 .43 .41 .13	Na2SiO3 06 .20 68 .70 .20 .14 .09	H <sub>2</sub> SO <sub>4</sub> 09 .22 71 .70 .22	Ca0 .04 .13 .59 .61 .13 .02 .10	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58 .57 .18 0 .11	KCI .14 .30 42 .43 .31	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29	MgSO <sub>4</sub> .09 .40 .31 .31 .41 .26 .32	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38 .39	TiO 0.5 .42 .16 .43 .43 .41 .38
V <sub>max</sub> I <sub>max</sub> phi Cos(phi) 1 <sup>st</sup> max 3 <sup>rd</sup> max 5 <sup>th</sup> max 7 <sup>th</sup> 7 <sup>th</sup>	7% .56 .90 .28 .27 .90 .85	NaCl 03 .13 43 .41 .13 .07	Na2SiO3 06 .20 68 .70 .20 .14	H <sub>2</sub> SO <sub>4</sub> 09 22 71 .70 .22 .16	Ca0 .04 .13 .59 .61 .13 .02	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58 .57 .18 0	KCI .14 .30 .42 .43 .31 .13	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19	MgSO <sub>4</sub> .09 .40 .31 .31 .41 .26	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38	TiO 0.5 .42 .16 .43 .43
$V_{max}$ $I_{max}$ phi Cos(phi) $1_{max}^{rd}$ $3_{max}^{rd}$ $5_{max}^{rb}$ $7_{max}^{rb}$ $1_{P}^{rb}$	7% .56 .90 .28 .27 .90 .85 .89	NaCl .03 .13 .43 .41 .13 .07 .07	Na2SiO3 06 .20 68 .70 .20 .14 .09	H <sub>2</sub> SO <sub>4</sub> 09 .22 71 .70 .22 .16 .09	Ca0 .04 .13 .59 .61 .13 .02 .10	Na <sub>2</sub> SO <sub>4</sub> 02 .17 58 .57 .18 0 .11	KCI .14 .30 .42 .43 .31 .13 .26	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29	MgSO <sub>4</sub> .09 .40 .31 .31 .41 .26 .32	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38 .39	TiO 0.5 .42 .16 .43 .43 .41 .38
$V_{max}$ $I_{max}$ phi Cos(phi) $1_{max}^{st}$ $3_{max}^{st}$ $5_{max}^{sb}$ $7_{max}^{sb}$ $1_{P}^{st}$ $3_{P}^{sf}$	7% .56 .90 .28 .27 .90 .85 .89 .90	NaCl .03 .13 .43 .41 .13 .07 .07 .10	Na2SiO3 06 .20 68 .70 .20 .14 .09 .06	H₂SO₄ 09 .22 71 .70 .22 .16 .09 .05	CaO .04 .13 .59 .61 .13 .02 .10 .03	Na₂SO₄ 02 .17 58 .57 .18 0 .11 .01	KCI .14 .30 .42 .43 .31 .13 .26 .15	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23	MgSO <sub>4</sub> .09 .40 .31 .41 .26 .32 .28	FeCl <sub>3</sub> .07 .26 .26 .48 .38 .39 .35	Ti0 0.5 .42 .16 .43 .41 .38 .36 .57 .53
$V_{max}$ $I_{max}$ phi Cos(phi) $1_{max}^{rd}$ $3_{max}^{rd}$ $5_{max}^{rb}$ $1_{P}^{rd}$ $3_{P}^{rf}$ $3_{P}^{rf}$	7% .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97	NaCl 03 .13 43 .41 .13 .07 .07 .10 .12 .10 .09	Na2SiO3 06 .20 68 .70 .20 .14 .09 .06 .18 .16 .12	H <sub>2</sub> SO <sub>4</sub> 09 .22 71 .70 .22 .16 .09 .05 .21 .20 .14	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19	Na2SO4 02 .17 58 .57 .18 0 .11 .01 .25 .07 .22	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37	Mg504 .09 .40 .31 .31 .41 .26 .32 .28 .46 .30 .42	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38 .39 .35 .57 .43 .53	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57
$V_{max}$ $I_{ann}$ phi Cos(phi) $1^{rd}_{max}$ $3^{rd}_{max}$ $7^{rd}_{max}$ $1^{rf}_{P}$ $3^{rf}_{P}$ $5^{rh}_{P}$	7; .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1	NaCl 03 .13 43 .41 .13 .07 .10 .12 .10 .09 .10	Na2SiO1 06 .20 68 .70 .20 .14 .09 .06 .18 .16 .12 .07	H <sub>2</sub> SO <sub>4</sub> 09 .22 71 .70 .22 .16 .09 .05 .21 .20 .14 .07	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19 .06	Na2SO4 02 .17 58 .57 .18 0 .11 .01 .25 .07 .22 .11	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .38 .19 .29 .23 .41 .22 .37 .29	MgSO4 .09 .40 .31 .41 .26 .32 .28 .46 .30 .42 .36	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .38 .39 .35 .57 .43 .53 .45	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .57 .57
$ \begin{array}{l} \mathbf{V}_{max} \\ \mathbf{I}_{am} \\ \mathbf{phi} \\ \mathbf{Cos(phi)} \\ 1_{max}^{rt} \\ 3_{max}^{rt} \\ 3_{max}^{rt} \\ 5_{max}^{tt} \\ 1_{P}^{rt} \\ 3_{P}^{rd} \\ 3_{p}^{rd} \\ 5_{p}^{tb} \\ \mathbf{Ncl} \end{array} $	7; .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1 .10	NaCl 03 .13 43 .41 .13 .07 .10 .12 .10 .09 .10 1	Na2SiO3 06 06 68 70 20 14 09 06 18 16 12 07 67	H_\$004 09 .22 71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19 .06 .42	Na2SO4 02 77 58 57 18 0 11 01 25 07 22 11 35	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24	MgSO4 .09 .40 -31 .31 .26 .32 .28 .46 .30 .42 .36 .19	FeCl <sub>3</sub> .07 .47 .26 .48 .38 .39 .35 .57 .43 .53 .45 .15	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10
$\begin{array}{c} \mathbf{V}_{max}\\ \mathbf{I}_{max}\\ \mathbf{phi}\\ \mathbf{Cos(phi)}\\ 1_{max}^{cl}\\ 3_{max}^{cl}\\ 3_{max}^{cl}\\ 1_{p}^{cl}\\ 3_{p}^{cl}\\ 3_{p}^{cl}\\ 3_{p}^{cl}\\ 3_{p}^{cl}\\ \mathbf{Nacli}\\ \mathbf{Nacsio,}\\ \mathbf{Nacsio,}\\ \end{array}$	7; .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1 .10 .07	NaCl 03 .13 43 .41 .13 .07 .07 .10 .12 .10 .09 .10 1 .67	Na2SiO3 06 .20 68 .70 .20 .14 .09 .06 .18 .16 .12 .07 .67 1	H_SO4 09 .22 71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19 .06 .42 .62	Na2SO4 02 7 58 57 18 0 11 01 25 07 22 11 35 52	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36	Mg504 .09 .30 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29	FeCl <sub>3</sub> .07 .47 .26 .48 .38 .39 .35 .57 .43 .53 .45 .15 .22	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10 .15
$\begin{array}{c} \mathbf{V}_{max}\\ \mathbf{I}_{aan}\\ \mathbf{phi}\\ \mathbf{Cos(phi)}\\ \mathbf{S}_{max}^{sd}\\ \mathbf{S}_{max}^{sd}\\ \mathbf{S}_{max}^{sd}\\ \mathbf{S}_{p}^{sd}\\ \mathbf{S}_{p}^{sd}\\ \mathbf{S}_{p}^{sd}\\ \mathbf{S}_{p}^{sd}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cs}_{cslo_{s}}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cs}_{cslo_{s}}\\ \mathbf{M}_{sol}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cslo_{s}}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cslo_{s}}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cslo_{s}}\\ \mathbf{Na}_{cl}\\ \mathbf{Na}_{cslo_{s}}\\ \mathbf{Na}_{csl}\\ \mathbf{Na}_{csl}\\$	7; .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .07	NaCl 03 .13 43 .41 .13 .07 .07 .10 .12 .10 .09 .10 1 .67 .52	Na2SiO3 06 06 68 .70 .20 .14 .09 .06 .18 .16 .12 .07 .67 1 .77	H_SO4 09 .22 71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77 1	Ca0 .04 .13 .59 .61 .13 .02 .10 .02 .02 .19 .06 .42 .62 .81	Na2SO4 02 7 58 .57 .18 0 .11 .01 .25 .07 .22 .11 .35 .52 .67	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43 .56	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46	Mg504 .09 .40 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38	FeCl <sub>3</sub> .07 .47 .26 .26 .48 .39 .35 .57 .43 .53 .43 .45 .15 .22 .29	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10 .15 .19
$ \begin{array}{l} V_{max} \\ V_{max} \\ I_{mm} \\ phi \\ Cos(phi) \\ 1_{max}^{rd} \\ 3_{max}^{rd} \\ 5_{max}^{rd} \\ 1_{p}^{rd} \\ 5_{p}^{rd} \\ 5_{p}^{rd} \\ 5_{p}^{rd} \\ 5_{p}^{rd} \\ NaCI \\ Na_{s}SiO_{4} \\ CaO \end{array} $	7% .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .07 .06	NaCl 03 .13 .41 .13 .07 .07 .10 .12 .10 .12 .10 .09 .10 1 .52 .42	Na2SiO3 06 06 68 68 70 .20 14 .09 .06 18 16 12 .07 .67 1 77 .62	H_SO4 09 .22 .71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77 1 .81	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19 .06 .42 .62 .81 1	Na2SO4 02 7 58 57 18 0 11 25 07 22 11 35 52 67 83	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43 .56 .69	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46 .57	Mg504 .09 .40 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38 .46	FeCl <sub>5</sub> .07 .26 .26 .48 .39 .35 .57 .43 .53 .43 .53 .45 .15 .22 .29 .36	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .57 .57 .57 .10 .15 .19 .24
$ \begin{array}{l} V_{uax} \\ I_{uax} \\ phi \\ Cos(phi) \\ 1_{max}^{rd} \\ 3_{max}^{rd} \\ 3_{max}^{rd} \\ 5_{p}^{ad} \\ 1_{p}^{rd} \\ 3_{p}^{rd} \\ 5_{p}^{b} \\ 7_{p}^{b} \\ Na_{c}SiO_{s} \\ H_{s}SO_{4} \\ CaO \\ Na_{s}SO_{4} \end{array} $	7% .56 .90 .28 .27 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .06 .11	NaCl 03 .13 .43 .41 .13 .07 .07 .10 .12 .10 .09 .10 1 .67 .52 .35	Na2SiO3 06 06 68 70 20 14 09 06 18 16 12 07 67 1 77 62 52	H_SO4 09 .22 71 .70 .22 .16 .09 .05 .21 .07 .20 .14 .07 .52 .77 1 .81 .67	Ca0 .04 .13 .59 .61 .13 .02 .10 .03 .02 .10 .03 .02 .19 .06 .42 .62 .81 1 .83	Na2SO4 02 17 58 .57 .18 0 .11 .01 .25 .07 .22 .11 .35 .52 .67 .83 .1	KCI .14 .30 .42 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43 .56 .69 .83	Al <sub>2</sub> SO <sub>4</sub> 111 .37 .43 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46 .57 .69	Mg504 .09 .40 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38 .46 .56	FeCl <sub>5</sub> .07 .26 .26 .48 .39 .35 .57 .43 .53 .45 .15 .22 .29 .36 .43	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10 .15 .19 .29
	7% .56 .90 .28 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .07 .07 .07 .21	NaCl 03 .13 43 .41 .13 .07 .10 .10 .10 .10 .10 .10 .10 .10 .10 .10	Na_SiO <sub>3</sub> -06 -06 -08 -70 -00 -14 -09 -06 -18 -16 -12 -07 -67 -1 -77 -62 -52 -43	H_SO4 09 .22 .71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77 1 .81 .67 .56	CaO .04 .13 .59 .61 .13 .02 .10 .03 .20 .02 .19 .06 .42 .62 .81 1 .83 .69	Na2SO4 02 58 57 58 57 18 0 11 01 25 07 22 11 35 52 67 83 1 1.83	KCI .14 .30 .42 .43 .31 .13 .26 .15 .32 .21 .29 .43 .56 .69 .83 1	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46 .57 .69 .83	Mg504 .09 .31 .31 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38 .46 .56 .67	FeCl <sub>5</sub> .07 26 .26 .26 .48 .38 .39 .35 .57 .43 .53 .45 .15 .22 .29 .36 .43 .52	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .57 .57 .10 .15 .19 .24 .29 .35
$ \begin{array}{l} V_{max} \\ I_{max} \\ phi \\ Cos(phi) \\ 1_{max}^{sr} \\ 3_{rd}^{sr} \\ 5_{max}^{sr} \\ 7_{p}^{sh} \\ 1_{r}^{r} \\ 3_{r}^{rd} \\ 5_{p}^{sh} \\ 7_{p}^{sh} \\ 7_{p}^{sh} \\ 1_{SO_4}^{sr} \\ CaO \\ Na_5SO_4 \\ KCI \\ Na_5SO_4 \\ KCI \\ Na_5SO_4 \\ \end{array} $	7; .56 .90 .22 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .07 .07 .01 .21 .29	NaCl 03 .13 .41 .13 .07 .07 .10 .10 .10 .10 .10 .10 .10 .10 .10 .10	Na_SiO, 06 06 68 70 68 70 20 14 09 06 18 16 12 07 67 1 76 52 52 36	H <sub>2</sub> SO <sub>4</sub> 09 71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77 1 .81 .67 .56 .46	Ca0 .04 .13 .59 .61 .13 .02 .02 .02 .02 .02 .02 .02 .02 .02 .02	Na2SO4 02 17 58 .57 18 0 .11 .01 .25 .07 .22 .07 .22 .07 .52 .52 .67 .83 1 .83 .69	KCI .14 .30 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43 .56 .69 .83 .83	Al <sub>2</sub> SO <sub>4</sub> 111 .43 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46 .57 .69 .83 .1	Mg504 .09 .31 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38 .46 .56 .67 .81	FeCl <sub>5</sub> .07 .26 .26 .26 .38 .39 .35 .57 .43 .53 .45 .15 .22 .36 .43 .52 .62	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10 .15 .19 .24 .29 .35 .42
$ \begin{array}{l} V_{nax} \\ I_{am} \\ phi \\ Cos(phi) \\ 1_{max}^{m} \\ 3_{max}^{rd} \\ 5_{max}^{sh} \\ 7_{p}^{sd} \\ 7_{p}^{sd} \\ 3_{p}^{rd} \\ 5_{p}^{sh} \\ 7_{p}^{sh} \\ Na_{2}SO_{4} \\ CaO_{4} \\ SO_{4} \\ CaO_{5} \\ KCI \\ Al_{5}SO_{4} \\ KCI \\ KCI \\ Al_{5}SO_{4} \\ KCI \\ K$	7; .56 .90 .27 .90 .85 .89 .90 .95 .88 .97 1 .07 .07 .06 .11 .21 .29 .36	NaCl 03 .13 .41 .13 .07 .07 .10 .12 .10 .10 .10 1 .10 1 .10 1 .52 .42 .35 .29 .24 .19	Na_SiO <sub>1</sub> 06 06 68 70 20 14 09 06 18 16 12 07 18 16 12 07 1 77 62 52 43 36 29	H_SO4 09 .22 .71 .70 .22 .16 .09 .05 .21 .20 .14 .52 .77 1 .52 .77 1 .67 .56 .46 .38	Ca0 .04 .13 .59 .61 .13 .02 .00 .02 .10 .02 .10 .02 .19 .06 .42 .81 1 .83 .69 .57 .46	Na2SO4 02 17 58 .57 18 0 .11 .01 .25 .07 .22 .11 .35 .52 .67 .83 1 .83 .69 .56	KCI .14 .30 .42 .43 .13 .26 .15 .32 .21 .43 .56 .69 .83 1 .83 .67	Al <sub>2</sub> SO <sub>4</sub> .11 .37 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .37 .29 .24 .37 .29 .24 .36 .46 .57 .83 .1 .81	Mg504 .09 .40 .31 .31 .41 .26 .32 .28 .46 .30 .42 .30 .42 .36 .29 .38 .46 .56 .56 .56 .57 .81	FeCl <sub>5</sub> .07 .26 .26 .26 .48 .39 .35 .57 .43 .53 .43 .53 .45 .15 .22 .29 .36 .43 .52 .62 .77	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .57 .57 .10 .15 .29 .35 .42 .52
$ \begin{array}{l} V_{max} \\ I_{max} \\ phi \\ Cos(phi) \\ 1_{max}^{sr} \\ 3_{rd}^{sr} \\ 5_{max}^{sr} \\ 7_{p}^{sh} \\ 1_{r}^{r} \\ 3_{r}^{rd} \\ 5_{p}^{sh} \\ 7_{p}^{sh} \\ 7_{p}^{sh} \\ 1_{SO_4}^{sr} \\ CaO \\ Na_5SO_4 \\ KCI \\ Na_5SO_4 \\ KCI \\ Na_5SO_4 \\ \end{array} $	7; .56 .90 .22 .90 .85 .89 .90 .95 .88 .97 1 .10 .07 .07 .01 .11 .21 .29	NaCl 03 .13 .41 .13 .07 .07 .10 .10 .10 .10 .10 .10 .10 .10 .10 .10	Na_SiO, 06 06 68 70 68 70 20 14 09 06 18 16 12 07 67 1 76 52 52 36	H <sub>2</sub> SO <sub>4</sub> 09 71 .70 .22 .16 .09 .05 .21 .20 .14 .07 .52 .77 1 .81 .67 .56 .46	Ca0 .04 .13 .59 .61 .13 .02 .02 .02 .02 .02 .02 .02 .02 .02 .02	Na2SO4 02 17 58 .57 18 0 .11 .01 .25 .07 .22 .07 .22 .07 .52 .52 .67 .83 1 .83 .69	KCI .14 .30 .43 .31 .13 .26 .15 .33 .15 .32 .21 .29 .43 .56 .69 .83 .83	Al <sub>2</sub> SO <sub>4</sub> 111 .43 .43 .43 .38 .19 .29 .23 .41 .22 .37 .29 .24 .36 .46 .57 .69 .83 .1	Mg504 .09 .31 .31 .31 .41 .26 .32 .28 .46 .30 .42 .36 .19 .29 .38 .46 .56 .67 .81	FeCl <sub>5</sub> .07 .26 .26 .26 .38 .39 .35 .57 .43 .53 .45 .15 .22 .36 .43 .52 .62	TiO 0.5 .42 .16 .43 .41 .38 .36 .57 .53 .57 .57 .10 .15 .19 .24 .29 .35 .42

From correlation matrix above, it is shown that the different parameters those yield values 0.9 and above are maximum amplitude of leakage current with maximum first, third, fifth and seventh harmonics, and maximum power of leakage current of first, fifth and seventh. This means the maximum amplitude of leakage current influences very significantly to those parameters. Those parameters rise as the leakage current increase very considerably. However, power of third harmonics is influenced by 83% of leakage current. It is significant. Whereas, the chemical compounds of H<sub>2</sub>SO<sub>4</sub> and Na<sub>2</sub>SiO<sub>3</sub> are affected to the power factor of 70%. This is significantly enough. Base on this correlation matrix, among these chemical compounds, both are the most conductive, because the power factor drastically increase. It means the conductivity increase more rapidly, rather than those due to other chemical compounds. Otherwise, applied voltage amplitude contributes only 69%, to the amplitude of leakage current. In other word, although the leakage current is almost influenced by the applied voltage, there are any other parameters for affecting the leakage current, i.e. in this

experiment is artificial pollution. The correlation components of other parameters are so small, so that their influential levels are not so significant.

## III. CONCLUSION

Base on the data and analyses, it can be concluded that the fundamental harmonic leakage current amplitude is significantly influenced by the applied voltage. The leakage current amplitude influences the parameters with maximum first, third, fifth and seventh harmonics, and maximum power of leakage current of first, fifth and seventh very considerably. The second one is for maximum power of third harmonic leakage current.

Whereas, the chemical compounds of  $H_2SO_4$  and  $Na_2SiO_3$  affected to the power factor around 70%. This is significantly enough. Among these chemical compounds, both are the most conductive, because the power factors drastically increase.

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