Irrigation Water Quality Evaluation in Jiaokou Irrigation District, Guanzhong Basin

Qiying Zhang, Panpan Xu, Hui Qian

Abstract—Groundwater is an important water resource in the world, especially in arid and semi-arid regions. In the present study, 141 groundwater samples were collected and analyzed for various physicochemical parameters to assess the irrigation water quality using six indicators (sodium percentage (Na%), sodium adsorption ratio (SAR), magnesium hazard (MH), residual sodium carbonate (RSC), permeability index (PI), and potential salinity (PS)). The results show that the patterns for the average cation and anion concentrations were in decreasing orders of Na⁺ > Mg²⁺ > Ca²⁺ > K⁺ and $SO_4^{2-} > HCO_3^{-} > Cl^{-} > NO_3^{-} > CO_3^{2-} > F^{-}$, respectively. The values of Na%, MH, and PS show that most of the groundwater samples are not suitable for irrigation. The same conclusion is drawn from the USSL and Wilcox diagrams. PS values indicate that Cl and SO₄²⁻have a great influence on irrigation water in Jiaokou Irrigation District. RSC and PI values indicate that more than half of groundwater samples are suitable for irrigation. The finding is beneficial for the policymakers for future water management schemes to achieve a sustainable development goal.

Keywords—Irrigation water quality evaluation, groundwater chemistry, Jiaokou Irrigation District, Guanzhong Basin.

I. INTRODUCTION

GROUNDWATER plays an important role for domestic drinking, industrial production, and agricultural irrigation all over the world, especially in arid and semi-arid regions where surface water and precipitation are usually scarce [1], [2]. The quality of groundwater has been regarded as a decisive factor for a country's sustainable development [3]. Agricultural irrigation requires much higher water quality than industrial water and even household water [2].

Especially for irrigated areas, the variation in groundwater chemistry has a strong effect on plants and soil, potentially damaging plants and reducing crop yields [4], [5]. Specifically, the physical effect of ions is to decrease the osmotic pressure in the structural cells of a plant, preventing water from reaching the branches and leaves [5]. Especially for irrigated areas located in arid and semi-arid areas, groundwater salinization is a common problem [6], [7], where the level of human activities may have significantly increased or disturbed the extent of these processes. This in turn results in crop reduction and subsequent negative impact on the economy and human society. This research has important reference value for regions and countries with food security

Qiying Zhang*, Panpan Xu, and Hui Qian are with the School of Water and Environment, Chang'an University, Xi'an 710054, Shaanxi, China and with the Key Laboratory of Subsurface Hydrology and Ecological Effects in Arid Region of the Ministry of Education, Chang'an University, Xi'an 710054, Shaanxi, China (*corresponding author: e-mail: zhangqiying@chd.edu.cn).

issues caused by irrigation water quality [8]-[10]. Therefore, it is a prerequisite to understand groundwater quality for irrigation purposes.

The aim of this study is to evaluate the groundwater suitability for irrigation purposes using different indices (sodium percentage (Na%), SAR, RSC, MH, PI, and PS) and to provide information for policymakers to achieve sustainable development.

II. STUDY AREA

The Jiaokou Irrigation District (34°30′7″-34°52′37″N, 109°12′40″–110°10′1″E), with 60 years of irrigation history, is one of the artificial irrigation areas within the Guanzhong Basin [11]. It is surrounded by water on three sides, with the Shichuan River in the west, Luo River in the east, and Wei River in the south and is located in the east of the Guanzhong Basin (Fig. 1). The climate is classified as warm temperate and semi-arid monsoon with a mean annual temperature of 13.4 °C, a precipitation of 548.5 mm, and annual evaporation of 1003.1 mm. The irrigation water comes from the Wei River with the Canal Head Station of the irrigation area located on the north bank of the river 2.5 km east of Jiaokou Town, Lintong District, Xi'an City (Fig. 1), and the volume of water taken from the river per year is 3.07×10^8 m³. The canal system is mainly distributed throughout the central and western parts of the irrigation area, which is dominated by cultivated land. However, the dry land is mainly developed in the sandbelt area in the east of the irrigation area. In addition, orchards and residential areas are scattered throughout the study area.

III. MATERIALS AND METHODS

A. Sample Collection and Laboratory Analysis

A total of 141 phreatic groundwater samples were collected in 2014 from the borewells/hand pumps in the Jiaokou Irrigation District (Fig. 1). The pH and TDS of groundwater samples were measured using a portable device in the field. Ca^{2^+} , Mg^{2^+} , $\text{CO}_3^{2^-}$, and HCO_3^- were measured by titration with a detection limit of 1.0 mg/L, whereas K^+ and Na^+ were measured using flame atomic absorption spectrometry. CI^- and $\text{SO}_4^{2^-}$ were determined by ion chromatography. To ensure the appropriate accuracies in the analyses, the ionic-balance-error for the ions in the water samples was within the acceptable limit of $\pm 5\%$ [1], [2], [11].

B. Evaluation Method

Six indicators, sodium percentage (Na%), SAR, RSC, MH, PI, and PS, were calculated to evaluate the status of the groundwater for irrigation purposes [2], [5], [12], [13]. All the

parameters used in the following equations were expressed in the milli-equivalent unit.

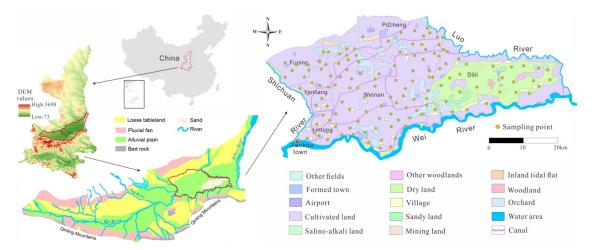


Fig. 1 Map showing the location and land use of study area as well as the distribution of sampling points

$$Na\% = \frac{Na^+ + K^+}{Na^+ + K^+ + Ca^{2+} + Ma^{2+}} \times 100$$
 (1)

$$SAR = \frac{Na^{+}}{\sqrt{(Ca^{2} + Mg^{2+})/2}}$$
 (2)

$$RSC = (C0_3^{2-} + HCO_3^{-}) - (Ca^{2+} + Mg^{2+})$$
 (3)

$$MH = \frac{Mg^{2+}}{(Ca^{2+} + Mg^{2+})} \times 100 \tag{4}$$

$$PI = \frac{Na^{+} + \sqrt{HCO_{3}^{-}}}{Na^{+} + K^{+} + Ca^{2+} + Mg^{2+}} \times 100$$
 (5)

$$PS = Cl^{-} + \frac{1}{2} \times SO_4^{2-}$$
 (6)

IV. RESULTS AND DISCUSSION

A. Groundwater Parameters and Groundwater Chemistry

The chemical characteristics of groundwater in this study region are presented in Table I. Seven out of 141 groundwater samples, located west of the study area, have a pH outside the recommended range of 6.5 to 8.5. TDS represents the total dissolved salts in water, and ranged from 237 to 7667 mg/L, with an average value of 2215 mg/L. The patterns for the average cation and anion concentrations were in decreasing orders of Na⁺> Mg²⁺> Ca²⁺> K⁺ and SO4²⁻> HCO³⁻> Cl⁻> NO³⁻> CO₃²⁻> F –, respectively (Table I). Specifically, the Na⁺ content ranged from 23 to 2822 mg/L (mean = 557.43 mg/L) in all groundwater samples, with 83.69% of samples exceeding the allowable limits of 200 mg/L, while the K⁺ concentration ranged from 0.37 to 96.46 mg/L, with an average value of 4.90 mg/L. Ca²⁺ and Mg²⁺ as important indicators of water hardness, showed values from 9.3 to 452.5 mg/L and 20.5 to 435.8 mg/L, respectively. The SO_4^{2-} concentrations ranged from 9.4 to 4367 mg/L, with an average value of 625.27 mg/L. Over 70% of samples exceeded the

threshold of 250 mg/L for SO₄²⁻. HCO³⁻ concentrations varied between 199 and 1249 mg/L, with a mean of 611.19 mg/L, while CO₃²⁻ values ranged between 11.6 and 120 mg/L, with an average value of 43.48 mg/L. In this study region, 85.82% of the groundwater samples exceeded the recommended levels of 300 mg/L for HCO³⁻. The Cl—concentrations varied between 3.16 and 1909 mg/L, with a mean of 379.37 mg/L, with 54.61% of the samples exceeding the threshold of 250 mg/L. NO³⁻ concentrations ranged from 0.1 to 1087 mg/L, with a mean of 156.5 mg/L, while F—concentrations were between 0.2 and 8.92 mg/L, with a mean value of 2.06 mg/L.

TABLE I
STATISTICAL SUMMARY OF CHEMICAL COMPOSITION OF GROUNDWATER IN
THE STUDY REGION

| Indices | Min. | Max. | Mean | SPL (WHO) | Number of Samples Exceeding the SPL | |
|------------------------------|------|-------|---------|-----------|--|--|
| pН | 6.48 | 9.9 | 7.68 | 6.5-8.5 | 7 | |
| TDS | 237 | 7667 | 2215.01 | 1000 | 117 | |
| K^{+} | 0.37 | 96.46 | 4.90 | 10 | 12 | |
| Na^+ | 23 | 2822 | 557.43 | 200 | 118 | |
| Ca^{2+} | 9.3 | 452.5 | 62.10 | 200 | 3 | |
| Mg^{2+} | 20.5 | 435.8 | 119.78 | 200 | 22 | |
| CO_3^{2-} | 11.6 | 120 | 43.48 | - | - | |
| HCO ₃ | 199 | 1249 | 611.19 | 300 | 121 | |
| Cl- | 3.16 | 1901 | 379.37 | 250 | 77 | |
| SO_4^{2-} | 9.4 | 4367 | 625.27 | 250 | 104 | |
| NO_3 | 0.1 | 1087 | 156.50 | 50 | 99 | |
| F | 0.2 | 8.92 | 2.06 | 0.5-1.5 | 93 | |
| $\mathrm{COD}_{\mathrm{Mn}}$ | 0.24 | 4.35 | 0.81 | 3 | 1 | |

Notes: SPL is standard permissible limit value.

B. Groundwater Quality for Irrigation Purposes

The ion concentration in water will affect the osmotic pressure of water entering the plant cells, which will affect the transportation of water in the plant and finally disrupt plant metabolism [5], [14]. Thus water quality assessment plays an important role in irrigation. In this study, various irrigation

water quality indices, Na%, SAR, RSC, MH, PI, and PS were calculated, and the results are summarized in Table II. The

spatial distribution maps of these indices are plotted in Fig. 2

TABLE II

| IL | WQ | N | P | IL | WQ | N | P |
|----------|-------------|----|--------|-------|-----------------------------|-----|--------|
| Na% | | | | SAR | | | |
| < 20 | excellent | 10 | 7.09% | <10 | excellent | 69 | 48.94% |
| 20-40 | good | 13 | 9.22% | 10-18 | good | 59 | 41.84% |
| 40-60 | permissible | 28 | 19.86% | 18-26 | doubtable | 12 | 8.51% |
| 60-80 | doubtable | 72 | 51.06% | >26 | unsuitable | 1 | 0.71% |
| >80 | unsuitable | 18 | 12.77% | | | | |
| RSC | | | | MH | | | |
| <1.25 | good | 80 | 56.74% | < 50 | suitable | 13 | 9.22% |
| 1.25-2.5 | doubtable | 9 | 6.38% | >50 | unsuitable | 128 | 90.78% |
| >2.5 | unsuitable | 52 | 36.88% | | | | |
| PI | | | | PS | | | |
| <25% | unsuitable | 0 | 0 | <3 | excellent to good | 18 | 12.77% |
| 25-75 | moderate | 67 | 47.52% | 3-5 | good to injurious | 9 | 6.38% |
| >75% | suitable | 74 | 52.48% | >5 | injurious to unsatisfactory | 114 | 80.85% |

Notes: IL is indices levels; WQ is water quality; N is the number of groundwater samples; P is the percentage.

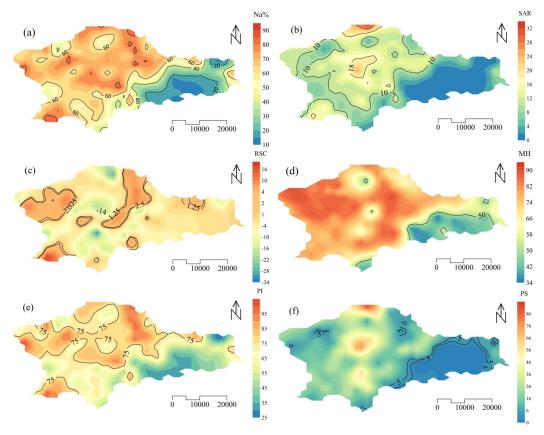


Fig. 2 The spatial distribution maps showing different irrigation indices

Na%, as an evaluation index of irrigation water quality, represents the sodium hazard. High Na% will damage the soil structure, reduce the permeability, and finally result in poor internal drainage in irrigation [12]. Na% is classified into five classes (Table II). The majority of the samples for irrigation are in the doubtable category (51.06%), followed by

permissible (19.86%), unsuitable (12.77%), good (9.22%), and excellent category (7.09%). Only 23 groundwater samples meet the regular irrigation. However, 72 and 18 samples have 60% to 80% and > 80% sodium, indicating that these samples are harmful for crops because of their effects on soil permeability and texture. As shown in Fig. 2 (a), except for a

small part of the eastern part of the study area, the Na% of the rest of the groundwater has permissible to unsuitable for irrigation.

SAR is used to assess the alkali/sodium level due to the excess sodium or limited calcium and magnesium [15]. Based on the SAR values, irrigation waters are classified into four categories (Table II). A total of 128 out of 141 groundwater samples are good for general irrigation because of limited calcium and magnesium. In addition, increasing the Ca²⁺ and Mg²⁺ concentration in water will improve soil permeability, and this is beneficial for groundwater samples that are not suitable for irrigation (less than 10%). From Fig. 2 (b), only a small part of the central and northern areas of the study area, the SAR values exceeded the specified value (> 18%).

The US Salinity Laboratory (USSL) diagram [16]; (Fig. 3 (a)) and Wilcox diagram [17]; (Fig. 3 (b)), combining both salinity hazard and alkalinity, were used further to analyze the

suitability of water for irrigation. Based on these data, most of the water samples fall into zone C3 and C4, indicating a high salinity hazard. Moreover, the SAR values increase with the increase of EC values (Fig. 3 (a)). In addition, as shown in the Wilcox diagram (Fig. 3 (b)), most of groundwater samples belong to the categories "permissible to doubtful" and "doubtful to unsuitable". Moreover, the groundwater samples fall into the zone of "permissible to doubtful" are mainly poor water. The fair groundwater is generally distributed in the category of "excellent to good", indicating that fair groundwater for drinking purposes is a good source for irrigation. Ten samples belong to the category "unsuitable" with EC more than 3000 µS/cm. For groundwater samples with TDS less than 1000, they are mainly distributed in the category of "excellent to good". It can be seen from Figs. 3 (a) and (b), the larger the TDS of the water, the less suitable it is for irrigation.

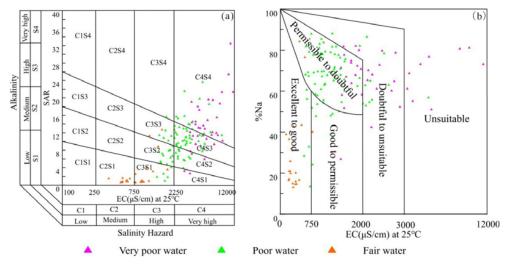


Fig. 3 USSL and Wilcox diagrams demonstrating irrigation water quality: (a) USSL diagram and (b) Wilcox diagram

RSC is defined as the difference of the sum of carbonate and bicarbonate and the sum of calcium and magnesium [2], [18]. In this study, the RSC values of 80 groundwater samples are less than 1.25, i.e., 56.74% of groundwater samples are good for irrigation purpose (Table II). However, 9 and 52 water samples belong to the doubtable and unsuitable category, respectively. Irrigation water with high RSC value may make the soil infertile due to the presence of sodium carbonate [19], [20]. Furthermore, the high-value area of RSC is small and scattered (Fig. 2 (c)). High bicarbonate concentration in irrigation water will raise the RSC values and increase water hardness due to the precipitation of calcium bicarbonate and magnesium bicarbonate [21].

Calcium and magnesium in groundwater generally maintain an equilibrium state. However, when more Mg²⁺ is present in groundwater, the crop yields will decrease due to the alkaline soil [14], [15]. Moreover, high level of magnesium in water can deteriorate soil structure because of exchangeable Na in irrigated soils. MH reflects the damage of magnesium to soil structure. The boundary value of MH for irrigation is 50 [22].

In the Jiaokou Irrigation District, 128 (90.78%) groundwater samples are unsuitable for irrigation based on the MH value (Table II) and these unsuitable samples are basically distributed throughout the study area (Fig. 2 (d)), confirming high magnesium content in the study area.

Long-term irrigation of the soil with mineral-rich (Ca²⁺, Mg²⁺, Na⁺, and HCO₃⁻) water will reduce soil permeability and eventually make soil to retard the emergence of seedlings [2], [4], [13], [15]. PI, as an indicator to reflect soil permeability, was proposed by [6] and classified into three categories, unsuitable (< 25%), moderate (25-75%), and suitable (> 75%). Some 47.52% of the groundwater samples belong to the moderate category and 52.48% belong to the suitable category (Table II). It can be seen that the impact of irrigation water in the study area is slight on soil permeability. The spatial distribution of PI is shown in Fig. 2 (e).

PS, as an indicator of the impact of Cl⁻ and SO₄²-on irrigation water, is defined as the Cl⁻ concentrations plus half of the SO₄²-concentration [2], [5]. The groundwater samples are classified into three categories, excellent to good, good to

injurious, and injurious to unsatisfactory, based on the PS values. The results show that 114 and 9 of groundwater samples belong to categories "injurious to unsatisfactory" and "good to injurious", respectively (Table II). However, only 18 of the samples belong to "excellent to good". This indicates that Cl^- and $\text{SO}_4{}^2$ —have a great influence on irrigation water. As shown in Figs. 2 (d) and (f), the spatial distribution of PS is basically consistent with MH.

V.CONCLUSIONS

Various physicochemical parameters of 141 groundwater samples were analyzed to assess the irrigation water quality using six indicators (Na%, SAR, RSC, MH, PI, and PS). The major conclusions of the study are as follows:

The patterns for the average cation and anion concentrations were in decreasing orders of $Na^+ > Mg^{2+} > Ca^{2+} > K^+$ and $SO_4^{2-} > HCO_3^- > Cl^- > NO_3^- > CO_3^{2-} > F^-$, respectively. The majority of the samples for irrigation based on Na% are in the doubtable category (51.06%), followed by permissible (19.86%), unsuitable (12.77%), good (9.22%), and excellent category (7.09%). The USSL and Wilcox diagrams show that most of groundwater samples are not suitable for irrigation. RSC values indicate 56.74% of groundwater samples are good for irrigation purpose, and 128 (90.78%) groundwater samples are unsuitable for irrigation based on the MH value and basically distributed throughout the study area. Based on the PI values, 47.52% of the groundwater samples belong to the moderate category and 52.48% are belong to the suitable category, and 114 and 9 of groundwater samples are belong to categories "injurious to unsatisfactory" and "good to injurious" based on the PS, indicating that Cl and SO₄² have a great influence on irrigation water. These findings would provide guidance for spatial management decisions of irrigation groundwater in Jiaokou Irrigation District.

ACKNOWLEDGMENTS

This study was financially supported by the National Natural Science Foundation of China (Grant No. 41572236, 41790441, and 41931285) and the Fundamental Research Funds for the Central Universities, CHD (No. 300102290401 and 300102290715). And the completion of this article was inseparable from the contributions of all authors. Their support is gratefully acknowledged.

REFERENCES

- Q.Y. Zhang, P.P. Xu, and H. Qian, "Groundwater Quality Assessment Using Improved Water Quality Index (WQI) and Human Health Risk (HHR) Evaluation in a Semi-arid Region of Northwest China," 2020, Expo Health.
- [2] P.P. Xu, W.W. Feng, H. Qian, and Q.Y. Zhang, "Hydrogeochemical Characterization and Irrigation Quality Assessment of Shallow Groundwater in the Central-Western Guanzhong Basin, China," *Int. J. Environ. Res. Public Health*, 2019, vol. 16, 1492.
- [3] K. Ravindra, P.S. Thind, S. Mor, T. Singh, and S. Mor, "Evaluation of groundwater contamination in Chandigarh: Source identification and health risk assessment," *Environ Pollut.* 2019, vol. 255: 113062.
- [4] J. Chen, Q.W. Huang, Y.L. Lin, Y. Fang, H. Qian, R.P. Liu, and H.Y. Ma, "Hydrogeochemical Characteristics and Quality Assessment of Groundwater in an Irrigated Region, Northwest China," Water, 2019,

- vol. 11(1): 18.
- [5] P. Tahmasebi, M.H. Mahmudy-Gharaie, F. Ghassemzadeh, and A. Karimi Karouyeh, "Assessment of groundwater suitability for irrigation in a gold mine surrounding area, NE Iran," *Environ. Earth Sci.*, 2018, vol. 77: 766
- [6] L.D. Doneen, "Notes on water quality in agriculture," Water Science and Engineering, *University of California*, Davis, 1964.
- [7] H. Jia, H. Qian, L. Zheng, W.W. Feng, H.K. Wang, and Y.Y. Gao, "Alterations to groundwater chemistry due to modern water transfer for irrigation over decades," *Sci. Total Environ*. 2020, vol. 717, 137170.
- [8] K.D. Brahman, T.G. Kazi, J.A. Baig, H.I. Afridi, A. Khan, S.S. Arain, and M.B. Arain, "Fluoride and arsenic exposure through water and grain crops in Nagarparkar, Pakistan," *Chemosphere*, 2014, vol. 100, pp.182-189
- [9] L. Karthikeyana, I. Chawla, A.K. Mishra, "A review of remote sensing applications in agriculture for food security: Crop growth and yield, irrigation, and crop losses," *J Hydrol*, 2020, vol. 586, 124905.
- [10] R.S. Kookana, P. Drechsel, P. Jamwal, J. Vanderzalm, "Urbanisation and emerging economies: Issues and potential solutions for water and food security," Sci. Total Environ., 2020, vol. 732, 139057.
- [11] Q.Y. Zhang, P.P. Xu, H. Qian, and F.X. Yang, "Hydrogeochemistry and Fluoride Contamination in Jiaokou Irrigation District, Central China: Assessment using multivariate statistical approach and human health risk," Sci. Total Environ. 2020, vol. 741, 140460.
- [12] N.S. Kawo, and K. Shankar, "Groundwater quality assessment using water quality index and GIS technique in Modjo River Basin, central Ethiopia," J. African Earth Sci., 2018, vol. 147, pp. 300-311
- [13] S.K. Khanoranga, "An assessment of groundwater quality for irrigation and drinking purposes around brick kilns in three districts of Balochistan province, Pakistan, through water quality index and multivariate statistical approaches," J. Geochem. Explor., 2018, vol. 11, 007.
- [14] P. Ravikumar, R.K. Somashekar, and M. Angami, "Hydrochemistry and evaluation of groundwater suitability for irrigation and drinking purposes in the Markandeya River basin, Belgaum District, Karnataka State, India," *Environ Monit Assess*, 2011, vol. 173(1), pp. 459-487
- [15] M. Kumar, K. Kumari, A.L. Ramanathan, and R. Saxena, "A comparative evaluation of groundwater suitability for irrigation and drinking purposes in two intensively cultivated districts of Punjab, India," *Environ Geol*, 2007, vol. 53, pp. 553-574.
- [16] United States Salinity Laboratory (USSL), "Diagnosis and improvement of saline and alkali soils," US Department of Agriculture (USDA), Washington, 1954, pp 69–81.
- [17] L.V. Wilcox, "The quality of water for irrigation use. US Department of Agriculture," Washington, 1948, pp. 1962.
- [18] S. Selvakumar, K. Ramkumar, N. Chandrasekar, N. Magesh, and S. Kaliraj, "Groundwater quality and its suitability for drinking and irrigational use in the Southern Tiruchirappalli district, Tamil Nadu, India," Appl. Water Sci., 2017, vol. 7, pp. 411-420.
- [19] P.Y. Li, Y.T. Zhang, N. Yang, L.J. Jing, and P.Y. Yu, "Major ion chemistry and quality assessment of groundwater in and around a mountainous tourist town of China," *Expo. Health*, 2016, vol. 8, pp. 239-252.
- [20] J.H. Wu, P.Y. Li, H. Qian, and Y. Fang, "Assessment of soil salinization based on a low-cost method and its influencing factors in a semi-arid agricultural area, northwest China," *Environ. Earth Sci.* 2014, vol. 71, pp. 3465-3475.
- [21] F.B. Owoyemi, G.E. Oteze, and O.V. Omonona "Spatial patterns, geochemical evolution and quality of groundwater in Delta State, Niger Delta, Nigeria: implication for groundwater management," *Environ Monit Assess*, 2019, vol. 191, 617.
- [22] K. Kalaivanan, B. Gurugnanam, H.R. Pourghasemi, M. Suresh, and S. Kumaravel, "Spatial assessment of groundwater quality using water quality index and hydrochemical indices in the Kodavanar subbasin, Tamil Nadu, India." Sustain Water Resour Manag. 2018, vol. 4, pp. 627–641.



Qiying Zhang was born in Pengyang County, Ningxia Hui Autonomous Region in May 1993. Zhang is a PhD candidate in Water Conservancy Engineering, School of Water and Environment, Chang'an University, Shaanxi Province, China. Zhang's main majors are hydrogeochemistry as well as hydrogeology and engineering geology. Panpan Xu is a PhD candidate in Water Conservancy Engineering, School of Water and

International Journal of Earth, Energy and Environmental Sciences

ISSN: 2517-942X Vol:15, No:1, 2021

Environment, Chang'an University. Xu's main major is hydrogeology and engineering geology. Hui Qian is professor in School of Water and Environment, Chang'an University. Qian is an authoritative expert in hydrogeochemistry.

She published 4 SCI papers in related fields, and the journals included in the papers are "Science of the Total Environment", "Exposure and Health", "International Journal of Environmental Research and Public Health", and "Arabian Journal of Geosciences".

Dr. Zhang was awarded a national scholarship during his master's and

Dr. Zhang was awarded a national scholarship during his master's and doctoral period.