

# The Effect of Particle Porosity in Mixed Matrix Membrane Permeation Models

Z. Sadeghi, M. R. Omidkhah, M. E. Masoumi

**Abstract**—The purpose of this paper is to examine gas transport behavior of mixed matrix membranes (MMMs) combined with porous particles. Main existing models are categorized in two main groups; two-phase (ideal contact) and three-phase (non-ideal contact). A new coefficient,  $J$ , was obtained to express equations for estimating effect of the particle porosity in two-phase and three-phase models. Modified models evaluates with existing models and experimental data using Matlab software. Comparison of gas permeability of proposed modified models with existing models in different MMMs shows a better prediction of gas permeability in MMMs.

**Keywords**—Mixed Matrix Membrane, Permeation Models, Porous particles, Porosity.

## I. INTRODUCTION

IN the recent years, membrane-based gas separation is one of the challenging industries in the world. The main membrane-based separations are  $H_2/CO_2$  separation for hydrogen production in fuel cells,  $CO_2/N_2$  separation in flue gas or lime oven exhaust gases,  $CO_2/CH_4$  separation for natural gas treatment or for biogas upgrading, and  $O_2/N_2$  separation for production oxygen enriched air or pure nitrogen. Membranes are categorized based on their structure, material, modules, which indicate that material category is important. Membranes are fabricated by different materials such as polymer, ceramic, carbon, metal, and liquid [1]-[8].

Different kinds of membranes were studied for gas separation, but polymeric membranes are the most common types used for gas separation, due to proper mechanical stability, processing capability, ease of operation and importantly economical cost [1]-[6].

The main criteria of polymeric membranes are selectivity and permeability in the membrane-based separation. In Fig. 1, as can be seen, the comparison of different kinds of membrane; in addition, some limitations were observed in trade-off between permeability and selectivity of polymeric membranes at Robesson graph [1], [3], [7], [8].

To overcome the problem of trade-off between permeability and selectivity, inorganic tiny fillers dispersed in polymeric membranes were applied to improve properties of polymeric membranes. This new membrane called Mixed Matrix

Membrane (MMM). Fig. 2 shows a schematic of mixed matrix membrane with different shape of particles [1]-[11].

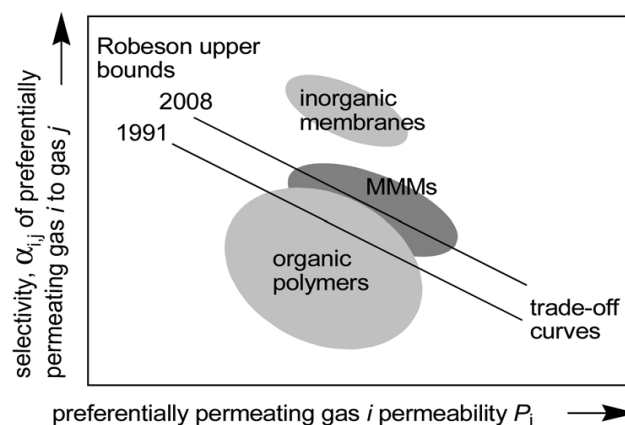


Fig. 1 Robeson graphs to compare different kinds of membranes for gas separation [9]

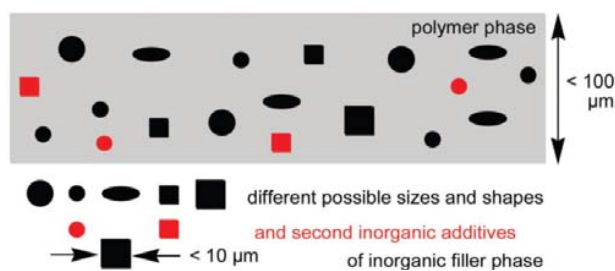


Fig. 2 Schematic of a mixed-matrix membrane [9]

MMMs are fabricated with different kinds of particles such as Carbon Molecular Sieve (CMS), activated carbon, silica, zeolites, nanoparticles and Metal Organic Framework (MOF) [12].

Regarding the literature [1]-[22], MMM models are categorized in two-phase and three-phase morphologies. Two-phase models are the first models in prediction of gas behavior, with assumption of ideal contact between particle & polymer and three-phase models are recommended based on weak interaction between particles and polymer matrix.

In this paper, the main existing permeation models of MMMs are reviewed. Then, the effect of porosity of particles in gas permeation through MMMs was studied. The porosity coefficient modifies existing models in two separate equations for two-phase and three-phase morphology. Gas relative permeability of modified models is validated with

Z. Sadeghi is with the Chemical Engineering Department, Islamic Azad University-Tehran North Branch, Tehran, P.O.Box.1667934783 Iran (phone: 982188770742; fax:+9821887872042; e-mail: z\_sadeghi@iau-tnb.ac.ir).

M. R. Omidkhah, Prof., was with UMIST, Manchester, UK. He is now with the Department of Chemical Engineering, Tarbiyat Modarres University, Tehran, P.O.Box 14115-143 Iran (e-mail: omidkhah@modares.ac.ir).

M. E. Masoumi is with the Chemical Engineering Department, Islamic Azad University-Tehran North Branch, Tehran, P.O.Box. 1667934783 Iran (e-mail: m\_masoumi@iau-tnb.ac.ir).

experimental data and calculated gas relative permeability of existing models by least square error and Matlab software.

## II. MODEL REVIEW

According to the literatures [1]-[6], [12], [16]-[21], there is a variety of permeation models for MMMs. As it was mentioned before, the main models categorized in two groups which are two-phase (particle-polymer) and three-phase (particle-interfacial layer-polymer) in Table I.

Two-phase models are based on ideal contact between polymer and dispersed phase. The two-phase models which

were considered are Maxwell, Bruggeman, Lewis-Nielsen, Pal, Chiew-Galandt, Bottcher, and Higuchi. In Three-phase models is considered a non-ideal contact and poor adhesion between particle and polymer. It can cause three defects; formation of a rigidified polymer layer around inorganic fillers, pore blockage in porous particles or creation of voids between polymer and particle. Therefore, it was assumed to consider an interfacial layer between polymer matrix and dispersed phase in three-phase models [1]-[6], [14]. Models of Modified Maxwell, Felske, Modified Felske and modified Pal are in categorization of three-phase models.

TABLE I  
SUMMARY OF MAIN EXISTING PERMEATION MODELS FOR MMMS [1]-[24]

Authors	Morphology	Base of model	Model
Maxwell [1]-[6], [14]	Two	Electrical conductivity	$P_r = \frac{1 + 2\phi(\lambda_d - 1)/(\lambda_d + 2)}{1 - \phi(\lambda_d - 1)/(\lambda_d + 2)}$
Bruggeman[1]-[6], [10],[13],[16]	Two	Thermal conductivity	$P_r^{1/3} \left( \frac{\lambda_{dm} - 1}{\lambda_{dm} + P_r} \right) = (1 - \phi)^{-1}$
Lewis- Nielsen [1]-[5], [18]	Two	Permeability	$P_r = \frac{1 + 2\phi(\lambda_d - 1)/(\lambda_d + 2)}{1 - \phi\psi(\lambda_d - 1)/(\lambda_d + 2)}$
Pal [1]-[5]	Two	Thermal conductivity	$\left( \frac{P_{eff}}{P_c} \right)^{1/3} \left[ \frac{\left( \frac{P_d}{P_c} \right) - 1}{\left( \frac{P_d}{P_c} \right) - \left( \frac{P_{eff}}{P_c} \right)} \right] = \left( 1 - \frac{\phi}{\phi_m} \right)^{-\phi_m}$
Bottcher[1],[5],[11],[8]	Two	Permeability	$\left( 1 - \frac{P_c}{P_{eff}} \right) (\alpha + 2 \frac{P_{eff}}{P_c}) = 3\phi(\alpha - 1)$
Chiew and Glandts [19],[20]	Two	Extension of Maxwell model	$\frac{P_{eff}}{P_c} = 1 + 3\beta\phi + K\phi^2 + O(\phi^3)$
Modified Maxwell [14],[23]	Three	Electrical conductivity	$P_r = \frac{P}{P_m} = \frac{2(1 - \phi) + (1 + 2\phi) \left( \frac{P_{eff}}{P_m} \right)}{(2 + \phi) + (1 - \phi) \left( \frac{P_{eff}}{P_m} \right)}$
Felske [14],[21]	Three	Thermal conductivity	$P_{eff} = P_1 \frac{2(1 - \phi_s) + (1 + 2\phi_s) \left( \frac{P_d}{P_1} \right)}{(2 + \phi_s) + (1 - \phi_s) \left( \frac{P_d}{P_1} \right)}$
Modified Felske [1]-[3]	Three	Thermal conductivity	$P_r = \frac{P}{P_m} = \frac{2(1 - \phi) + (1 + 2\phi)(\beta/\gamma)}{(2 + \phi) + (1 - \phi)(\beta/\gamma)}$
Modified Pal [1],[3]	Three	Thermal conductivity	$P_r = \frac{1 + 2\phi(\beta - \gamma)/(\beta + 2\gamma)}{1 - \psi\phi(\beta - \gamma)/(\beta + 2\gamma)}$
			$(P_{eff}/P_m)^{1/3} \left( \frac{\lambda_{di} - 1}{\lambda_{di} - \left( \frac{P_{eff}}{P_m} \right)} \right) = \left( 1 - \frac{\phi_s}{\phi_m} \right)^{-\phi_m}$
			$P_r^{1/3} \left( \frac{\lambda_{eff-m} - 1}{\lambda_{eff-m} - (P_r)} \right) = \left( 1 - \frac{\phi_s}{\phi_m} \right)^{-\phi_m}$

## III. INVESTIGATION OF PARTICLE POROSITY IN MMM MODELS

MMMs fabricated of polymer matrix and inorganic particles for improvement polymeric membrane properties. Dispersed particles in polymeric matrix are categorized in two groups; porous and dense (non-porous) particles. In prediction of gas permeability in MMMs, Existing models has been proposed without considering particle porosity (J coefficient). J coefficient is a new factor which introduces to correct effect of loading particles in MMMs. According to the researches, MMMs fabricated with porous particles enhanced selectivity and permeability compare to the MMM with non-porous particles [1]. The reason is that porosity in the gas transport not only decreases resistance through membrane, but also increases separation or selectivity of the gas based on porosity of the particles in the matrix [8], [22], [24].

Molecular sieving is the basis of separation in MMMs with porous particles. It should be notified that the number of particle pores is effective to calculate real gas permeability through membrane.

J coefficient is studied to modify mathematical models for MMMs. J coefficient is defined based on particle porosity in

two-phase and three-phase models in (1) and (2) respectively. ( $\epsilon$  is porosity percentage of particles)

$$J = 2 - 1/\epsilon \quad (1)$$

$$J = 2 - \epsilon \quad (2)$$

In two-phase models, no defect is presumed in contact between particle and polymer. This assumption leads to gas permeability with higher error against three-phase models. This coefficient applies the effect of the interfacial layer in calculations. There is some superiority for modifications of the existing models by this method. One of the most important privileged criteria is that estimating experimental parameters such as interfacial layer permeability, thickness, chain immobilization factor, and permeability reduction factor are not essential to calculate gas permeability. However, measurements or estimating of these parameters in three-phase models are needed. But, usage of J coefficient in (1) helps to estimate precisely gas permeability in existing two-phase models. Fast and easy estimations in industrial applications are another advantage of modification by this method.

Based on three-phase morphology the effects of three defects shall be considered in three-phase modeling. These three defects are sieve in a cage, rigidified polymer layer and pore blockage. However, permeability of interfacial layer in several three-phase models included Felske and modified Felske, is estimated based on the worst cases, i.e. rigidification layer and pore blockage, and It is obvious the effect of sieve in a cage and leaky case is not investigated in these models. Therefore, gas permeability in mentioned existing three-phase models is calculated lower than experimental observations. In the modified models by applying the correction of filler loading percentage in (2), the effect of sieve in a cage and leakage in interphase considered.

Modified models are reported in Table II. In this table all the existing permeation models modified with J coefficient.

TABLE II  
IMPROVEMENT OF EXISTING MODELS WITH J COEFFICIENT

Model	Modified Models with J coefficient
Maxwell [1]-[6], [14]	$P_r = \frac{1 + 2\phi J(\lambda_{dm} - 1)/(\lambda_{dm} + 2)}{1 - \phi J(\lambda_{dm} - 1)/(\lambda_{dm} + 2)}$
Bruggeman [1]-[6], [10],[13],[16]	$P_r^{1/3} \left( \frac{\lambda_{dm} - 1}{\lambda_{dm} - P_r} \right) = (1 - \phi J)^{-1}$
Lewis- Nielsen [1]-[5], [18]	$P_r = \frac{1 + 2\phi J(\lambda_{dm} - 1)/(\lambda_{dm} + 2)}{1 - \phi J(\lambda_{dm} - 1)/(\lambda_{dm} + 2)}$
Pal [1]-[5]	$(P_r)^{1/3} \left[ \frac{\lambda_{dm} - 1}{\lambda_{dm} - P_r} \right] = \left(1 - \frac{\phi J}{\phi_m}\right)^{-\phi_m}$
Botcher [1],[5],[11],[8]	$\left(1 - \frac{P_c}{P_{eff}}\right)(\alpha + 2\frac{P_{eff}}{P_c}) = 3\phi J(\alpha - 1)$
Chiew & Glandts [19],[20]	$\frac{P_{eff}}{P_c} = 1 + 3\beta\phi J + K(\phi J)^2 + O(\phi J)^3$
Felske [14]	$P_r = \frac{P}{P_m} = \frac{2(1 - \phi J) + (1 + 2\phi J)(\beta/\gamma)}{(2 + \phi J) + (1 - \phi J)(\beta/\gamma)}$
Modified Felske [1]-[3]	$P_r = \frac{1 + 2\phi J(\beta - \gamma)/(\beta + 2\gamma)}{1 - \psi\phi J(\beta - \gamma)/(\beta + 2\gamma)}$
Modified Maxwell [14], [23]	<i>This model considered influence of three status of non-ideal contact morphology in modeling MMM. Therefore, J coefficient factor is not essential to be used.</i>
Modified Pal [1], [3]	<i>This model considered influence of three status of non-ideal contact morphology in modeling MMM. Therefore, J coefficient factor is not essential to be used.</i>

#### IV. VALIDATION

Validity of proposed Modified models has been evaluated by least square method and compared with experimental data and existing permeation models; Maxwell, Bruggeman, Lewis-Nielsen, Pal, Chiew-Galandt, Botcher, Felske and modified Felske. The experimental data of MMMs which are used in this paper [3] are Matrimid-5218 matrix filled with CMS for separation CO<sub>2</sub>/CH<sub>4</sub> in 0.17, 0.19, 0.33, 0.36 filler loading percentage, Matrimid-5218 filled with CMS for separation O<sub>2</sub> of O<sub>2</sub>/N<sub>2</sub> in 0.19, 0.33, 0.36 filler loading percentage, BAPD-BPADA filled with Zeolit4A for separation O<sub>2</sub> of O<sub>2</sub>/N<sub>2</sub> in 0.15, 0.2, 0.3, 0.4 filler loading percentage, PVAC filled with Zeolit4A for separation O<sub>2</sub> of O<sub>2</sub>/N<sub>2</sub>, in 0.15, 0.25, 0.4 filler loading percentage.

For an illustration for two-phase models, in Fig. 3, gas relative permeability of proposed modified model of Maxwell, Chiew-Galant, Lewis, Bruggman and Pal have been compared with existing models. As can be seen in Fig. 3, modified models with considering particle porosity have better

anticipation of gas relative permeability compare to existing models.

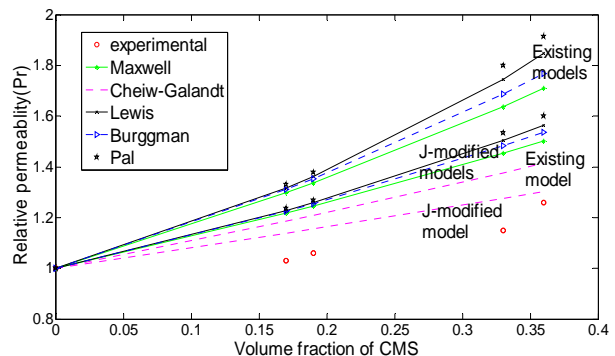


Fig. 3 Comparison of relative permeability CO<sub>2</sub> of CO<sub>2</sub>/CH<sub>4</sub> composition at Matrimid/CMS of existing two-phase models and J-modified models with 80% porosity

In Fig. 4, it can be seen an instance for three-phase model. The gas relative permeability of proposed three-phase model of Felske has been compared with the relative permeability of existing model. It can be observed, the modified models with considering particle porosity in their formula has a better prediction of gas relative permeability compare to existing models.

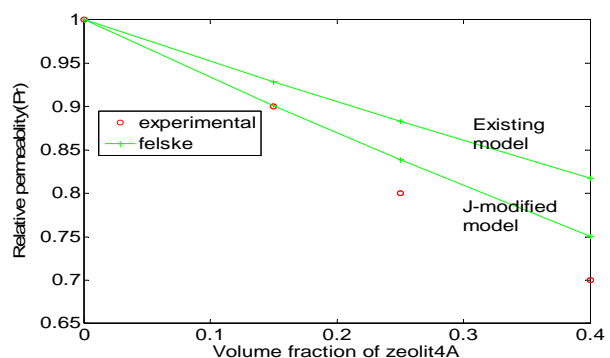


Fig. 4 Comparison of relative permeability O<sub>2</sub> of O<sub>2</sub>/N<sub>2</sub> composition at PVAC/Zeolit4A of existing three-phase models and J-modified models with 60% porosity

In Fig. 5, proposed modified model have been compared to two-phase and three-phase existing models by least square error values. To be more specific, among the modified models Chiew-Galandt and modified Felske are the best models for prediction of gas behavior through this MMM.

In Figs. 6-8 also proposed modified models were checked versus existing models and experimental data, the results are similar to the gas behavior observed in Fig. 5.

Overall, not only relative permeability error of modified models are dramatically less than existing two-phase models, but also three-phase models have a better prediction compare to two-phase models. Consequently, with considering the influence of effective porosity of particles in gas permeability

calculation of MMMs, the results of modified models are precisely closer to experimental data.

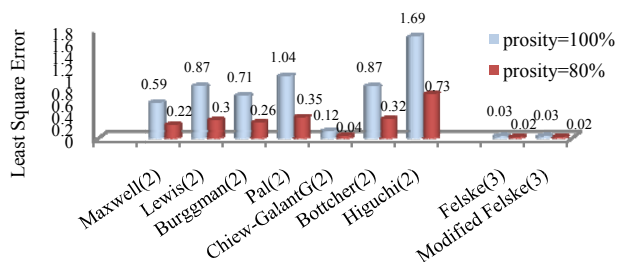


Fig. 5 Comparing error percentage of exist model with improved model with J coefficient for CO<sub>2</sub> separation of CO<sub>2</sub>/CH<sub>4</sub> in Matrimid/CMS

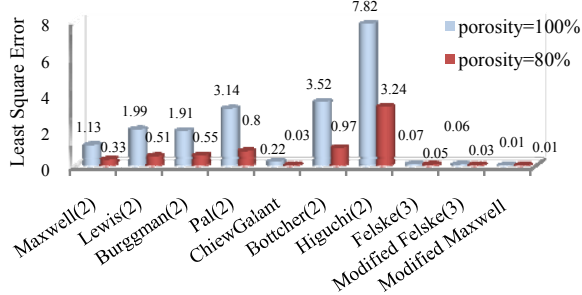


Fig. 6 Comparing error percentage of exist model with improved model with J coefficient for O<sub>2</sub> separation of O<sub>2</sub>/N<sub>2</sub> in Matrimid/CMS

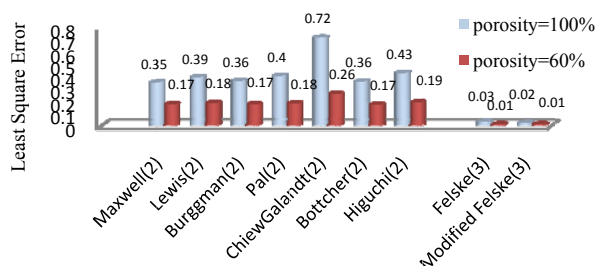


Fig. 7 Comparing error percentage of exist model with improved model with J coefficient for O<sub>2</sub> separation of O<sub>2</sub>/N<sub>2</sub> in BAPD-BPADA/Zeolit4A

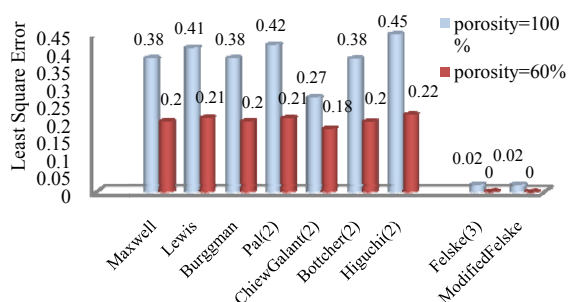


Fig. 8 Comparing error percentage of exist model with improved model with J coefficient for O<sub>2</sub> separation of O<sub>2</sub>/N<sub>2</sub> in PVAC/Zeolit4A

## V. CONCLUSION

In two-phase models, this coefficient considers simply effect of rigidification layer and pore blockage in estimation of gas permeability in MMMs and it doesn't need to assume interphase permeability, interfacial layer thickness and rigidification factor. Therefore, with this assumption, existing two-phase models improve easily and fast.

In three-phase models, gas permeability is calculated with considering the worst condition of fabrication in MMM (rigidification layer, pore blockage), therefore estimated permeability in existing three-phase models miss probability of sieve-in-cage or weak interaction between polymer and particle (void in MMMs). Thus, with correction effective pore percentage in filler loading by J coefficient, a more appropriate prediction is resulted in modified models.

In the final analysis, regarding the reasons mentioned above, it can be concluded that least square error of modified mathematical models with porosity coefficient is dramatically less than calculated error in existing models. In addition, three-phase models are in close agreement to experimental data. Therefore, it is proved that proposed modified models are nearby to experimental data and the results demonstrate a logical theory of gas behavior prediction.

## NOMENCLATURE

CMS	carbon molecular sieves
MMMs	mixed matrix membranes
$P_r$	relative permeability
$r$	radius of a spherical material
$R$	distance from the center of the sieve to boundary of the polymer

### Greek letters

$\alpha_r$	relative selectivity
$\beta$	called matrix rigidification or chain immobilization factor
$\delta$	the ratio of outer radius of rigidified interfacial matrix chain layer to core radius
$\psi$	Parameter described as function of packing volume fraction of filler particles
eff*	combined sieve and rigidified interfacial matrix chain layer polymer matrix
$\Phi$	the volume fraction of the filler particles
$\Phi_z$	combined volume fraction of the sieve phase and the interfacial rigidified matrix chains in the whole system.
$\Phi_d$	volume fraction of the dispersed phase
$\Phi_m$	maximum packing volume fraction of the dispersed phase
$\Phi_s$	volume fraction of the sieve phase in the combined sieve and rigidified interfacial matrix chain layer phase
$\gamma$	parameter described for ratio of the Interphase thickness to the particle radius
$\psi$	parameter described as function of packing volume fraction of filler particles
$\lambda$	permeability ratio

### Superscripts

cal	calculated
exp	experimental
NDP	number of data points

### Subscripts

d	dispersed phase
i	interphase

j counter  
 m polymer matrix  
 r relative  
 S inorganic phase in the combined inorganic and interphase phase  
 1,2 penetrant gas through membrane  
 c refer to permeability of a penetrant in the continuous phase  
 d refer to permeability of a penetrant in the disperse phase  
 eff refer to permeability in the composite membrane  
 l membrane thickness  
 P permeability  
 δ interphase thickness

## REFERENCES

- [1] H.Vinh-Thang, S.Kaliaguine "Predictive Models for Mixed-Matrix Membrane Performance: A Review", *ACS Publications J.*, vol.113 (7), 2012, pp. 4980–5028.
- [2] S.A.Hashemifard, A.F.Ismail, T.Matsuura, "A new theoretical gas permeability model using resistance modeling for mixed matrix membrane systems", *Membr. Sci.J.*, vol.350, 2010, pp. 259–268.
- [3] B.Shimekit, H.Mukhtar, T.Murugesan, "Prediction of the relative permeability of gases in mixed matrix membranes", *Membr. Sci.J.*, vol.373, 2011, pp 152–159.
- [4] K.M.Gheimasi, T.Mohammadi, O.Bakhtiari, "Modification of ideal MMMs permeation prediction models: Effects of partial pore blockage and polymer chain rigidification", *Membr. Sci.J.*, vol.427, 2013, pp 399–410.
- [5] E.E.Gonzo, M.L.Parentis, J.C.Gohifredi, "Estimating models for predicting effective permeability of mixed matrix membranes", *Membr. Sci.J.*, vol. 277, 2006, pp 46–54.
- [6] T.S.Chung, L.Y.Jiang, S.Kulprathipanja, "Mixed matrix membranes (MMMs) comprising organic polymers with dispersed in organic fillers for gas separation", *Prog.Polym.Sci. J.*, vol.32, 2007, pp. 483–507.
- [7] Z.Y.Yeo, T.L.Chew, P.W.Zhu, A.R.Mohamed, S.P.Chai, "Conventional processes and membrane technology for carbon dioxide removal from natural gas: A review", *Natural Gas Chemistry J.*, vol.21, 2012, pp.282–298.
- [8] J.H.Petropoulos, "A comparative study of approaches applied to the permeability of binary composite polymeric materials", *Polym. Sci.Polym. Phys. J.*, Ed. 23, 1985, pp. 1309–1324.
- [9] H.B.TanhJeazet, C.Staudt, C.Janiak, "Metal-organic frameworks in mixed-matrix membranes for gas separation", *Dalton Trans.J.*, vol.41, 2012, pp. 14003–14027.
- [10] D.A.G.Bruggeman, *Ann.phys. (Leipzig)*, vol.24, 1935, pp.636.
- [11] C.J.F.Bottcher, "The dielectric constant of crystalline powders", *Recueil desTravauxChimiques des Pays-Bas (Leipzig)*, vol.64, 1945, pp.47-51.
- [12] N.Mansouri, N.Rikhtegar, H.A.Panahi, F.Atabaki, B.Karimishahraki, "Porosity, characterization and structural properties of natural zeolite-clinoptilolite – as a sorbent", *Environment Protection EngineeringJ.*, vol. 39, No. 1, 2013, pp.139-153.
- [13] A.Bushell, "Mixed Matrix Membranes of a Polymer of Intrinsic Micro porosity with Crystalline Porous Solids", A thesis submitted to the University of Manchester, 2012.
- [14] R.Pal, "Permeation models for mixed matrix membranes", *Colloid and Interface Science J.*, vol. 317, 2008, pp.191–198.
- [15] M.A.Aroon, A.F.Ismail, T.Matsuura, M.M.Montazer-Rahmati, "Performance studies of mixed matrix membranes for gas separation: a review", *Sep. Purif. Technol.J.*, vol.75, 2010, pp.229–242.
- [16] A.Shariati, M.R.Omidkhal, M.Zamani, "New permeation models for nanocomposite polymeric membranes filled with nonporous particles", *Chem.Eng.Research and Design J.*, vol.90, 2012, pp.563-575.
- [17] W.I.Higuchi, T.Higuchi, "Theoretical analysis of diffusion at movement through heterogeneous barriers", *Amer. Pharm. Assoc., Sci. J.*, vol.49, 1960, pp.598-606.
- [18] L.E.Nielsen, "Thermal conductivity of particulate-filled polymers", *Appl.Polym.Sci. J.*, vol.17, 1973, pp. 3819.
- [19] E.E.Gonzo, "Estimating correlations for the effective thermal conductivity of granular materials", *Membr. Sci.J.*, vol.427, 2002, pp.399–410.
- [20] Y.C.Chiew, E.D.Glandt, "The effect of structure on the conductivity of a dispersion", *Colloid and Interface science J.*, vol.94, No.1, 1983, pp. 90-104.
- [21] J.D.Felske, "Effective thermal conductivity of composite spheres in a continuous medium with contact resistance", *Heat and Mass Transfer J.*, vol.47, 2004, pp.3453-3461.
- [22] M.Mohammadi, G.D.Najafpour, M.AbdulRahman, "Production of Carbon Molecular Sieves from Palm shell through carbon deposition from methane", *Chemical Industry & Chemical Engineering Quarterly J.*, vol.17 (4), 2011, pp.525-533.
- [23] D.Q.Vu, W.J.Koros, S.J.Miller, "Mixed matrix membranes using carbon molecular sieves II. Modeling permeation behavior", *Membr. Sci.J.*, vol.211, 2003, pp.335-348.
- [24] F.Dorosti, M.Omidkhal, R.Abedini, "Fabrication and characterization of Matrimid/MIL-53 mixed matrix membrane for CO<sub>2</sub>/CH<sub>4</sub> separation" *Chem.Eng.Research and Design J.*, to be published.

**Z. Sadeghi** was born in Tehran, Iran in 1979. She got her B.S. in chemical engineer from Isfahan University of Technology, Isfahan, Iran, 2003 and she is a student in last semester M.Sc. in chemical engineering of Islamic Azad University-Tehran North Branch, Tehran, Iran.

She is now a senior process engineer in TACECO (TarhAndishan Consulting Engineering Company in field of oil, gas and petrochemical projects) and a senior process engineer in Karoon Petrochemical Complex Plant. Her publications included "Modeling of mixed matrix membranes containing MOF-MIL-53 particles", Tehran, Iran, Civilika, 2014, "New Permeation model for mixed matrix membrane", Phuket, Thailand, IJCEA, 2014(Accepted). In addition, the main research areas are Process and Heat Exchanger simulation and modeling, Heat Exchanger Design.

Ms. Sadeghi became a Member of IACHE (Iran Association of Chemical Engineers).

**M.R. Omidkhal** was born in Tehran, Iran in 1958. He got his B.Sc., in chemical engineering in 1982 from Amir Kabir University of Technology, Tehran, Iran. Also he got M.Sc. in chemical engineering in 1985 from Wayne State University of Michigan, USA and Ph.D. in the same field in 1990 from UMIST, Manchester, UK.

He is now President of Chemistry and chemical engineering Research Center of Iran and a faculty member of chemical engineer department in TarbiatModarres University. The recent journal publications are: "Hydrogen separation and purification with poly (4-methyl-1-pentyne)/MIL 53 mixed matrix membrane based on reverse selectivity", Elsevier, 2014, In Press. "CO<sub>2</sub>/CH<sub>4</sub> separation by using poly methyl pentene/MIL 53 particles mixed matrix membrane.", Iranian Journal of Polymer Science & Technology, 2014, In Press. "Investigating the effect of dianhydride type and pyrolysis condition on the gas separation performance of membranes derived from blended polyimides through statistical analysis", Journal of Industrial and Engineering Chemistry, 2013, In Press. Also the main research areas are Process modeling and Control, Optimal Control of Thermal Cracking Reactors, Process Integration and Heat Exchanger Network Design. Process Synthesis Design and Optimization, Process Integration, Pinch Technology, Modeling and Simulation, Process Heat Recovery Systems, Energy Management.

Professor.Omidkhal is the member of IACHE (Iran Association of Chemical Engineers), Member of IPI (Iranian Petroleum Institute), Member of Iranian Gas Engineering Association, Member of Iranian Energy Association, Member of IIES (Institute for International Energy Studies), Member of IAEE (Iran Association for Energy Economics), Member of the board of IAEE, Member of the World Energy Council (WEC), National Energy, Committee of Iran, Member of the board of TPI (Technology Park of Iran), Chairman of the board of TPI (Technology Park of Iran), President of IACHE (Iranian Association of Chemical Engineering), Member of editorial board of Petroleum Research Journal, Member of editorial board of Farayandno Journal, Member of editorial board of Iranian Journal of Chemical Engineering, Member of editorial board of Iranian Energy Economics Journal, Member of editorial board of Iranian journal of Industry & Development, Advisor to the minister of Industry, Consultant to National Iranian Petrochemical Company (NIPC), Consultant to National Iranian Oil Company (NIOC).

**M.E. Masoumi** was born in shabestar, Iran in 1966. He got his B.Sc., in chemical engineering in 1990 from Sharif University of Technology in Tehran, Iran. Also he got M.Sc. and Ph.D. in the same field in 1994 and 2003 from TarbiatModarres University in Tehran, Iran.

He had two summer fellowship in Tabriz and Tehran refineries. Now he is a faculty member and head of chemical engineering department in Islamic Azad University – Tehran North Branch. The main journal publications are: "Simulation and Optimization of a Naphtha Thermal Cracking Pilot Plant.", Iranian Journal of Chemistry & Chemical Engineering, Vol. 22, No. 1, 2003, "Simulation, Optimization and Control of a Thermal Cracking Furnace.", Energy, Vol. 31, No.4, 2006 and "Modeling and Control of a Naphtha Thermal Cracking Pilot Plant." I&EC Research, Vol. 45, No.10 , 2006. Also the main research areas are Process modeling and Control, Optimal Control of Thermal Cracking Reactors, Process Integration and Heat Exchanger Network Design.

Dr. Masoumi is the member of IChE (Iran Association of Chemical Engineers) and IACSIT (International Association of Computer Science and Information Technology).